



# Química Inorgânica Avançada

## IQU-708

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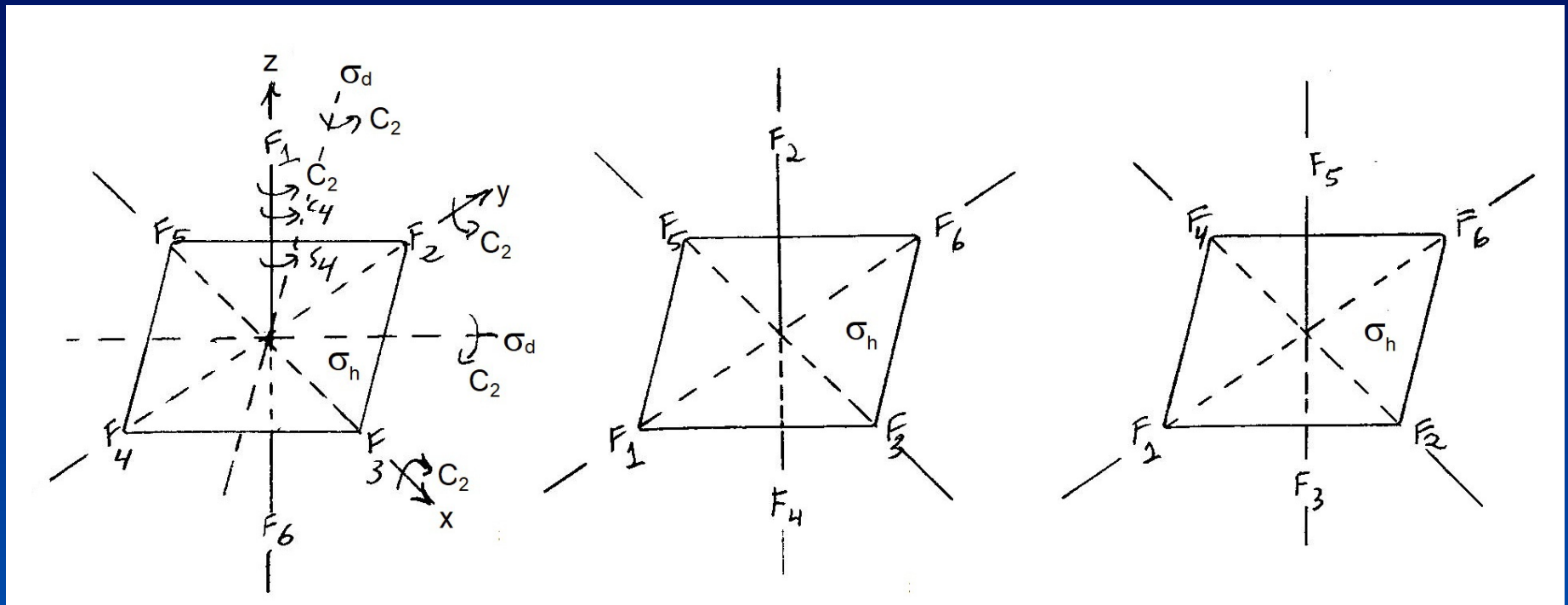
Simetria - Aula 5 - Geometria octaédrica  
Complexos com ligantes de campo  
fraco, forte e intermediário  
Transições eletrônicas

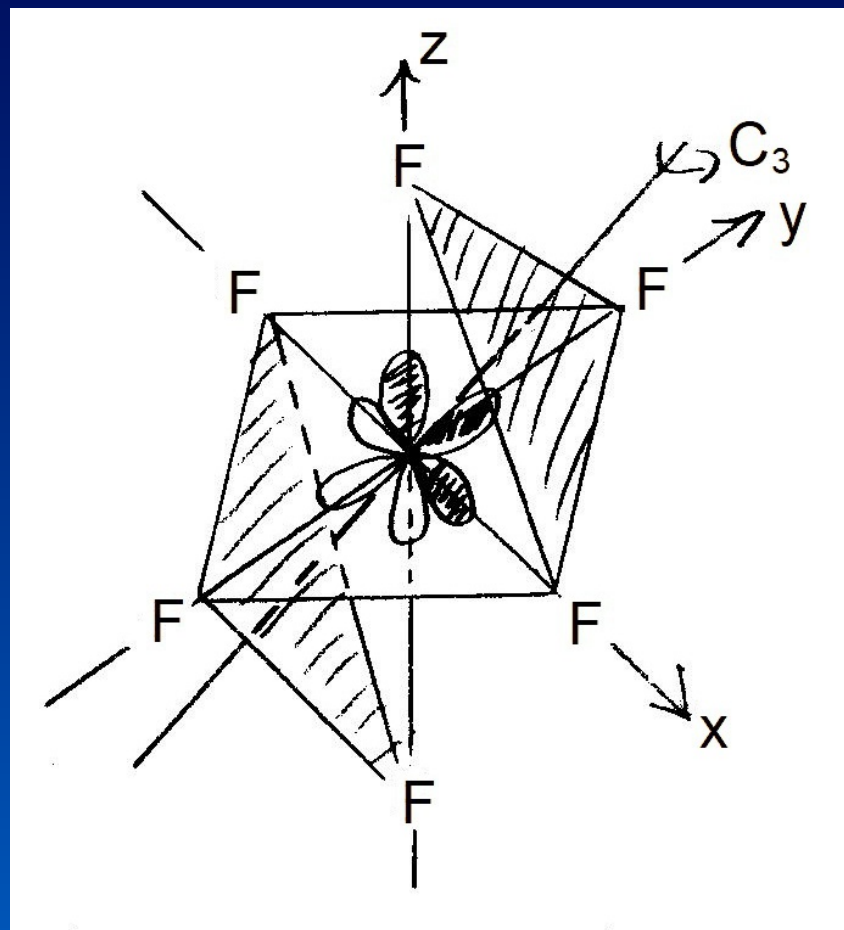
16/03/2024

# [CoF<sub>6</sub>]<sup>3-</sup> - O<sub>h</sub>

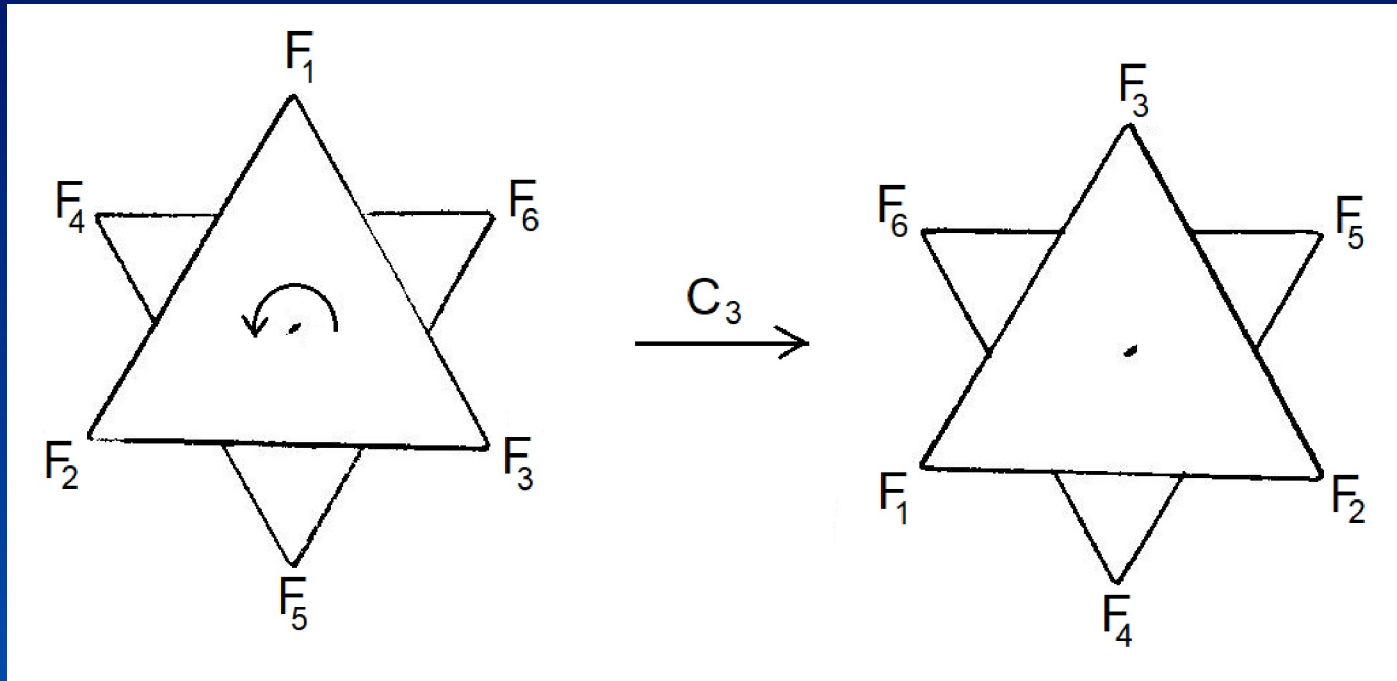
O <sub>h</sub>	E	8C <sub>3</sub>	6C <sub>2</sub>	6C <sub>4</sub>	3C <sub>2</sub> (=C <sub>4</sub> <sup>2</sup> ) (x,y,z)	i	6S <sub>4</sub>	8S <sub>6</sub>	3σ <sub>h</sub>	6σ <sub>d</sub>
A <sub>1g</sub>	1	1	1	1	1	1	1	1	1	1
A <sub>2g</sub>	1	1	-1	-1	1	1	-1	1	1	-1
E <sub>g</sub>	2	-1	0	0	2	2	0	-1	2	0
T <sub>1g</sub>	3	0	-1	1	-1	3	1	0	-1	-1
T <sub>2g</sub>	3	0	1	-1	-1	3	-1	0	-1	1
A <sub>1u</sub>	1	1	1	1	1	-1	-1	-1	-1	-1
A <sub>2u</sub>	1	1	-1	-1	1	-1	1	-1	-1	1
E <sub>u</sub>	2	-1	0	0	2	-2	0	1	-2	0
T <sub>1u</sub>	3	0	-1	1	-1	-3	-1	0	1	1
T <sub>2u</sub>	3	0	1	-1	-1	-3	1	0	1	-1

# $[\text{CoF}_6]^{3-} - \text{O}_h$

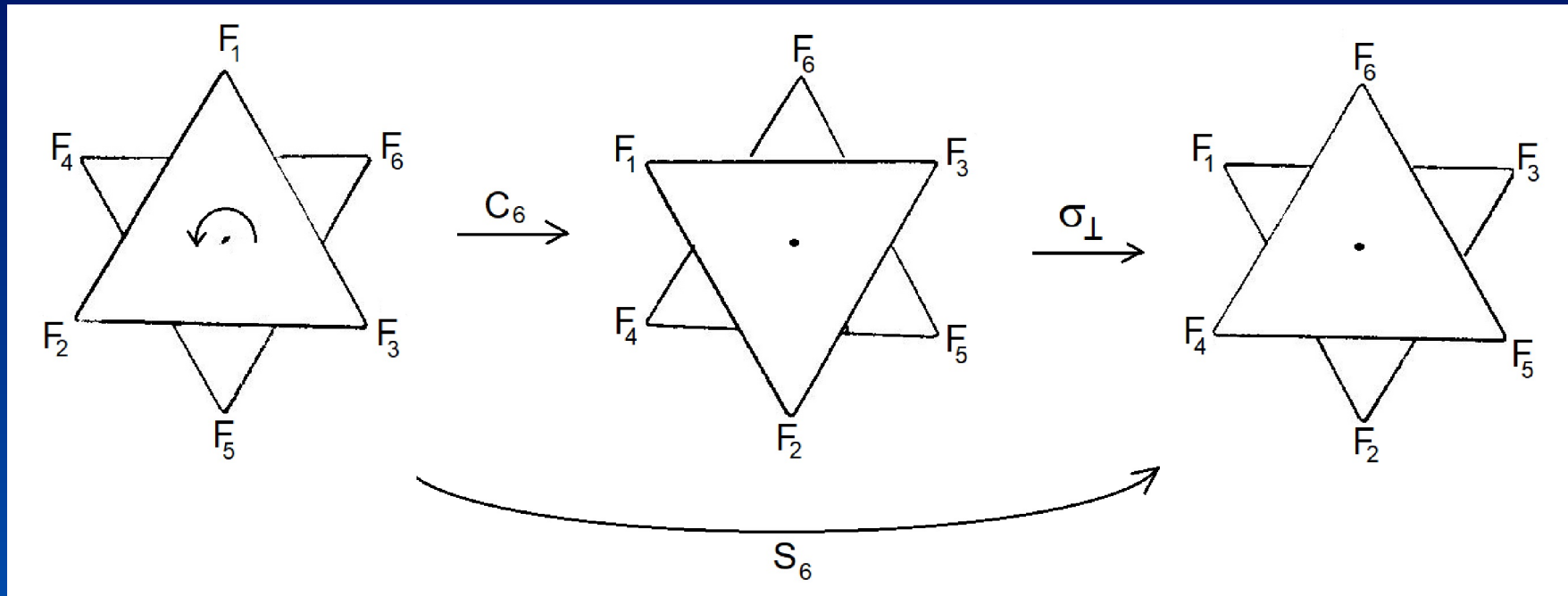




# $[\text{CoF}_6]^{3-} - \text{O}_h$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

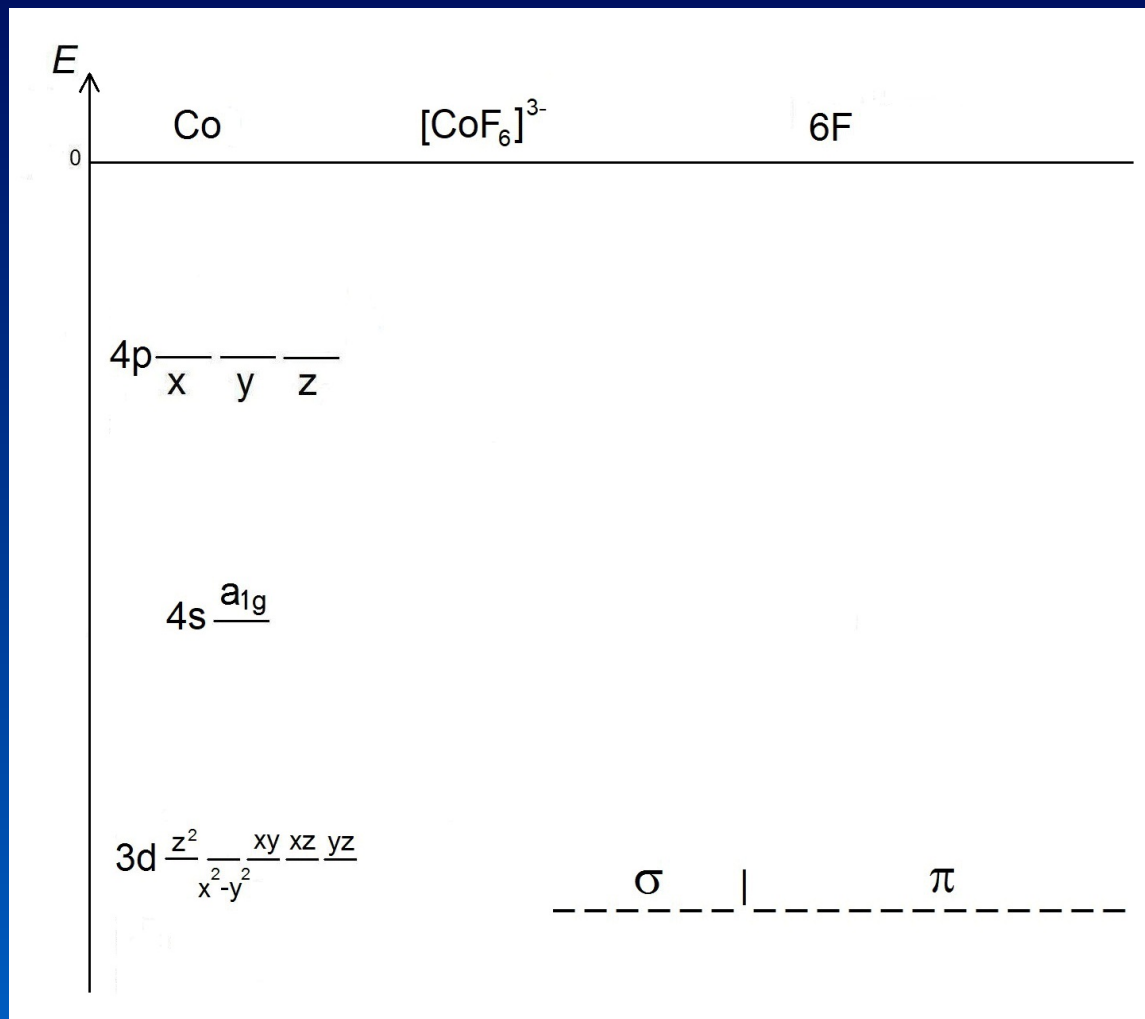


Classificando o orbital 4s do Co

		(x,y,z)										
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$		
$A_{1g}$	1	1	1	1	1	1	1	1	1	1	1	
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1	-1	
$E_g$	2	-1	0	0	2	2	0	-1	2	0	0	
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1	-1	
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1	1	
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1	-1	
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1	1	
$E_u$	2	-1	0	0	2	-2	0	1	-2	0	0	
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1	1	
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1	-1	
<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	<hr/>	
4s	1	1	1	1	1	1	1	1	1	1	1	$A_{1g}$

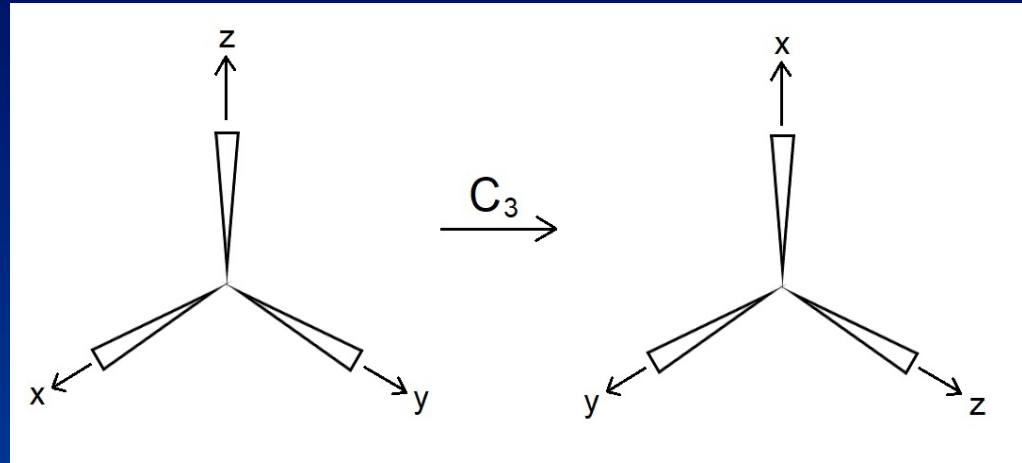
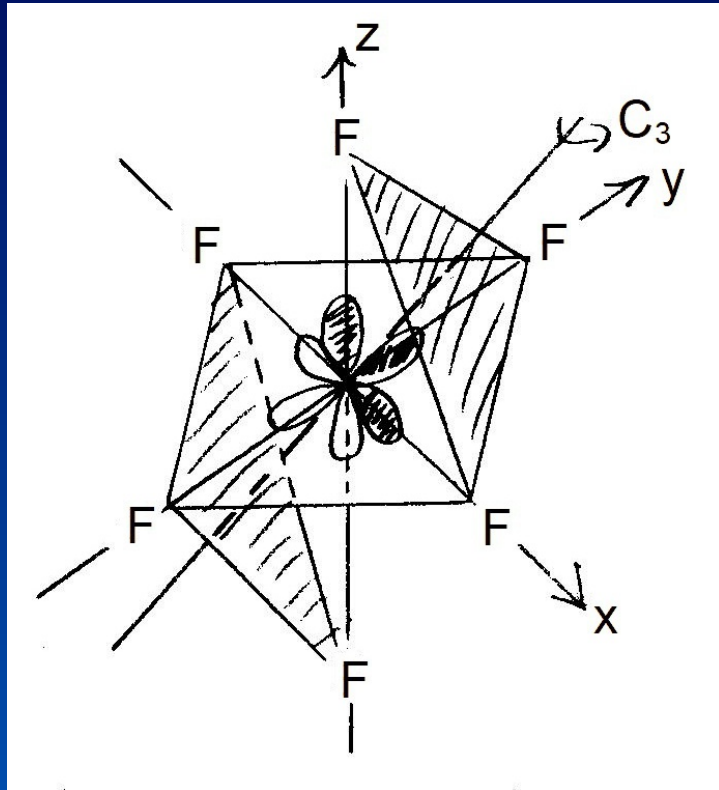
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Construindo o diagrama de energia dos orbitais moleculares





# Classificando os orbitais 4p do Co - INSEPARÁVEIS



$$\begin{aligned}x &-C_3 \rightarrow y \\y &-C_3 \rightarrow z \\z &-C_3 \rightarrow x\end{aligned}$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS

	$2p_x$	$2p_y$	$2p_z$			$2p_x$	$2p_y$	$2p_z$
$2p_x$	1	0	0	$C_3$	$2p_x$	0	1	0
$2p_y$	0	1	0	$\rightarrow$	$2p_y$	0	0	1
$2p_z$	0	0	1		$2p_z$	1	0	0

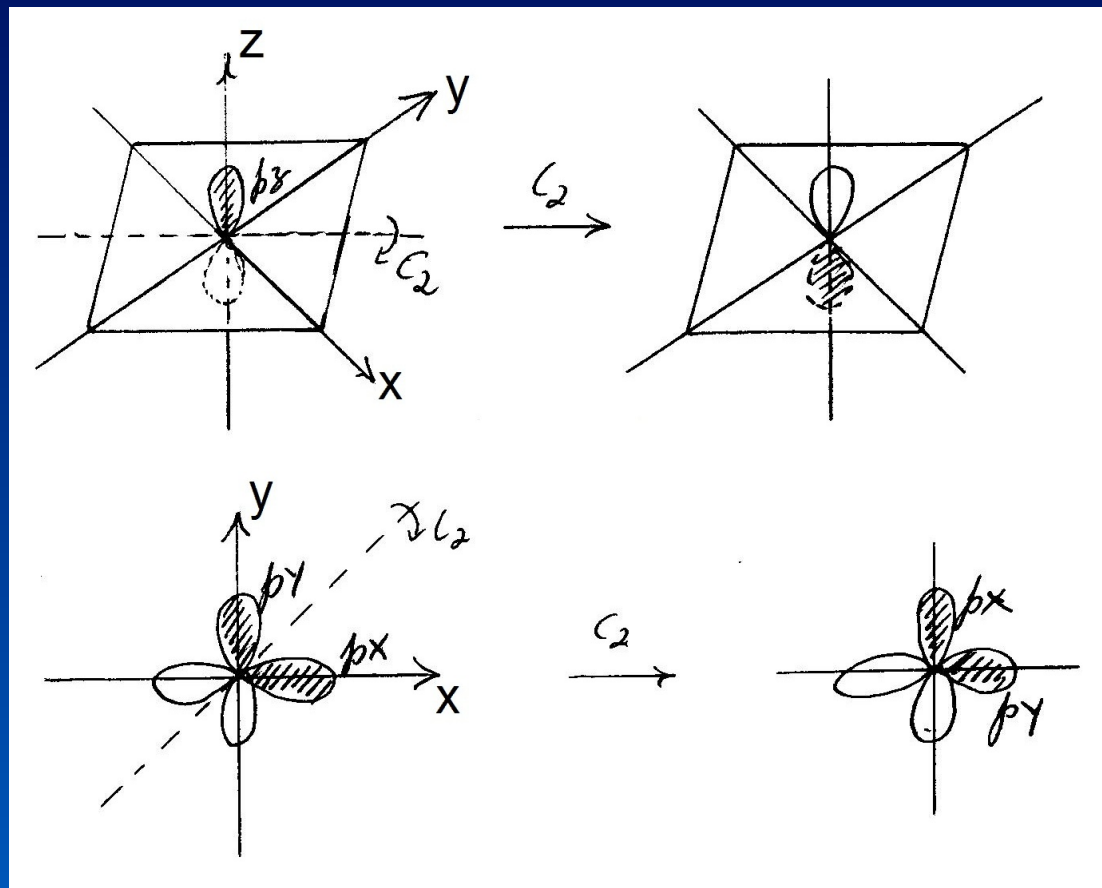
$$\chi = 0$$

## Classificando os orbitais 4p do Co - INSEPARÁVEIS

		(x,y,z)									
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	
$A_{1g}$	1	1	1	1	1	1	1	1	1	1	
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1	
$E_g$	2	-1	0	0	2	2	0	-1	2	0	
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1	
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1	
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1	
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1	
$E_u$	2	-1	0	0	2	-2	0	1	-2	0	
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1	
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1	
4p	3	0									

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS

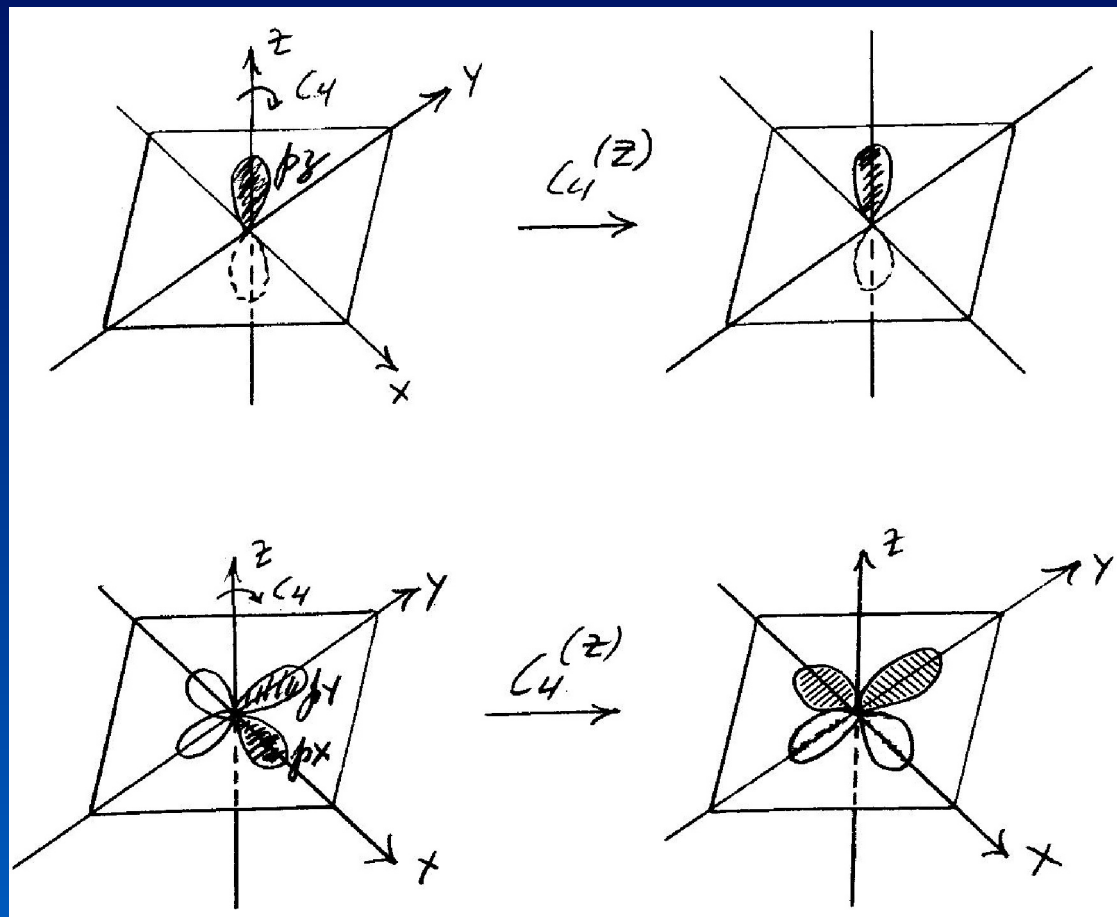
	$2p_x$	$2p_y$	$2p_z$		$2p_x$	$2p_y$	$2p_z$	
$2p_x$	1	0	0	$C_2$	$2p_x$	0	1	0
$2p_y$	0	1	0	$\rightarrow$	$2p_y$	1	0	0
$2p_z$	0	0	1		$2p_z$	0	0	-1

$$\chi = -1$$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS

	$2p_x$	$2p_y$	$2p_z$		$2p_x$	$2p_y$	$2p_z$	
$2p_x$	1	0	0	$C_4$	$2p_x$	0	1	0
$2p_y$	0	1	0	$\rightarrow$	$2p_y$	-1	0	0
$2p_z$	0	0	1		$2p_z$	0	0	1

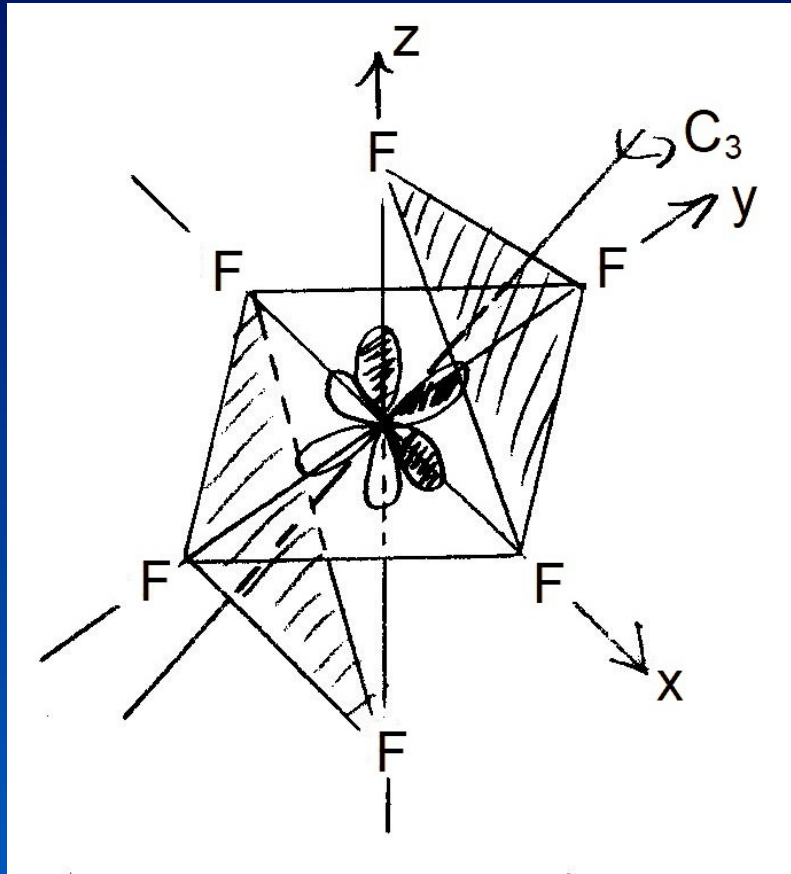
$$\chi = 1$$







Classificando os orbitais 4p do Co - INSEPARÁVEIS



i = centro de  
inversão

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais 4p do Co - INSEPARÁVEIS

	$2p_x$	$2p_y$	$2p_z$		$2p_x$	$2p_y$	$2p_z$	
$2p_x$	1	0	0	i	$2p_x$	-1	0	0
$2p_y$	0	1	0	→	$2p_y$	0	-1	0
$2p_z$	0	0	1		$2p_z$	0	0	-1

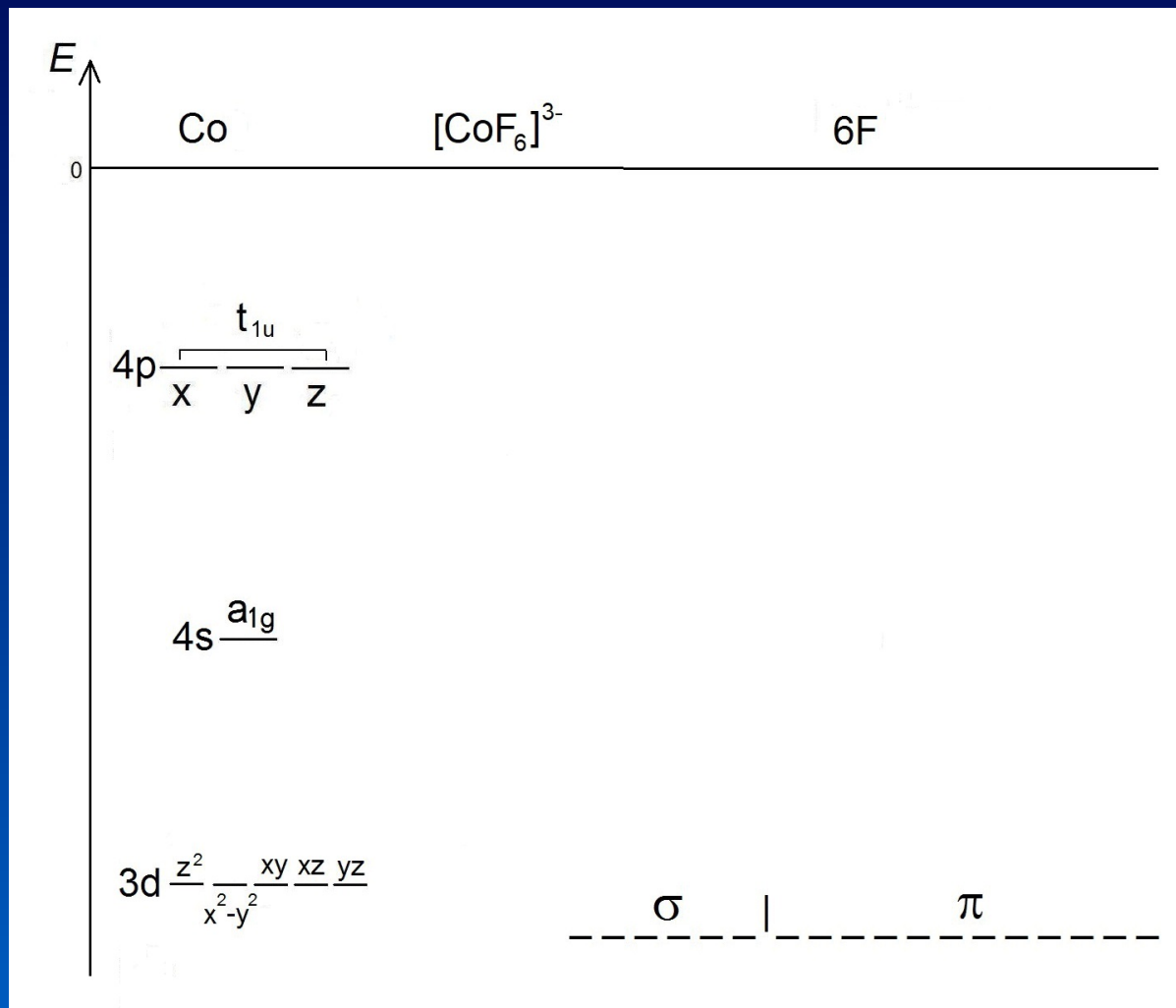
$$\chi = -3$$

## Classificando os orbitais 4p do Co - INSEPARÁVEIS

	E	8C <sub>3</sub>	6C <sub>2</sub>	6C <sub>4</sub>	3C <sub>2</sub> (=C <sub>4</sub> <sup>2</sup> ) (x,y,z)	i	6S <sub>4</sub>	8S <sub>6</sub>	3σ <sub>h</sub>	6σ <sub>d</sub>
O <sub>h</sub>	1	1	1	1	1	1	1	1	1	1
A <sub>1g</sub>	1	1	-1	-1	1	1	-1	1	1	-1
A <sub>2g</sub>	2	-1	0	0	2	2	0	-1	2	0
E <sub>g</sub>	3	0	-1	1	-1	3	1	0	-1	-1
T <sub>1g</sub>	3	0	1	-1	-1	3	-1	0	-1	1
T <sub>2g</sub>	1	1	1	1	1	-1	-1	-1	-1	-1
A <sub>1u</sub>	1	1	-1	-1	1	-1	1	-1	-1	1
A <sub>2u</sub>	2	-1	0	0	2	-2	0	1	-2	0
E <sub>u</sub>	3	0	-1	1	-1	-3	-1	0	1	1
T <sub>1u</sub>	3	0	1	-1	-1	-3	1	0	1	-1
T <sub>2u</sub>										
4p	3	0	-1	1		-3				T <sub>1u</sub>

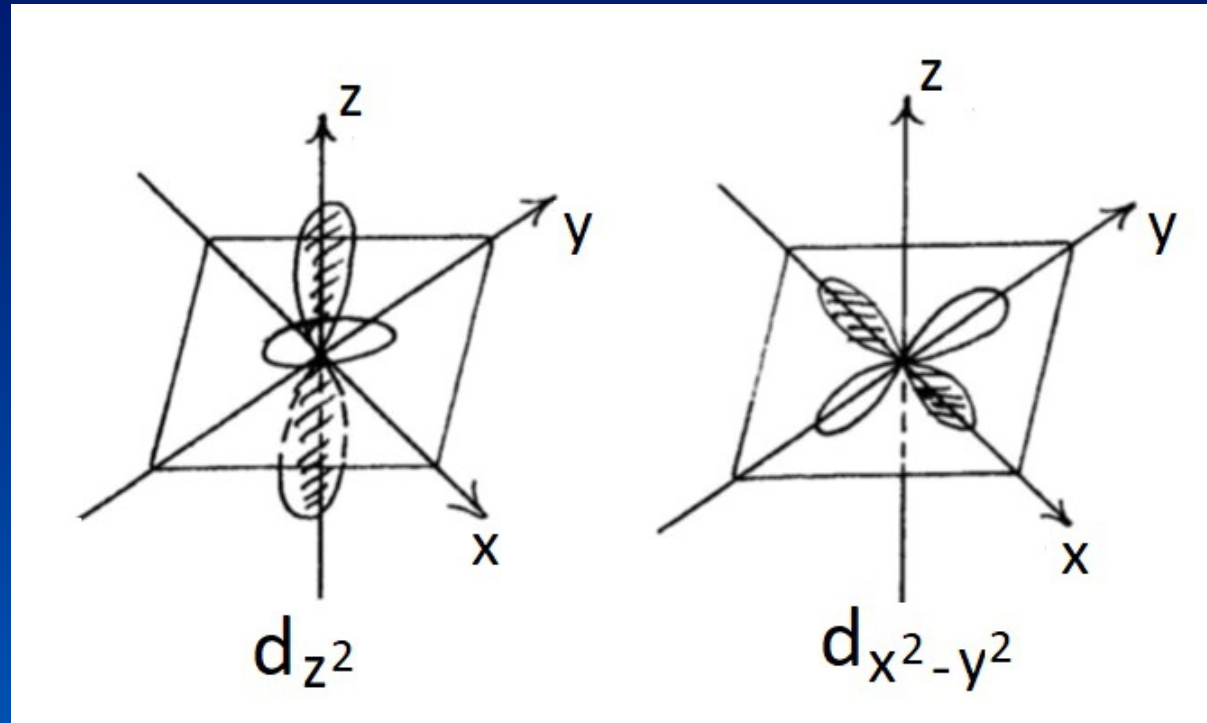
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Construindo o diagrama de energia dos orbitais moleculares



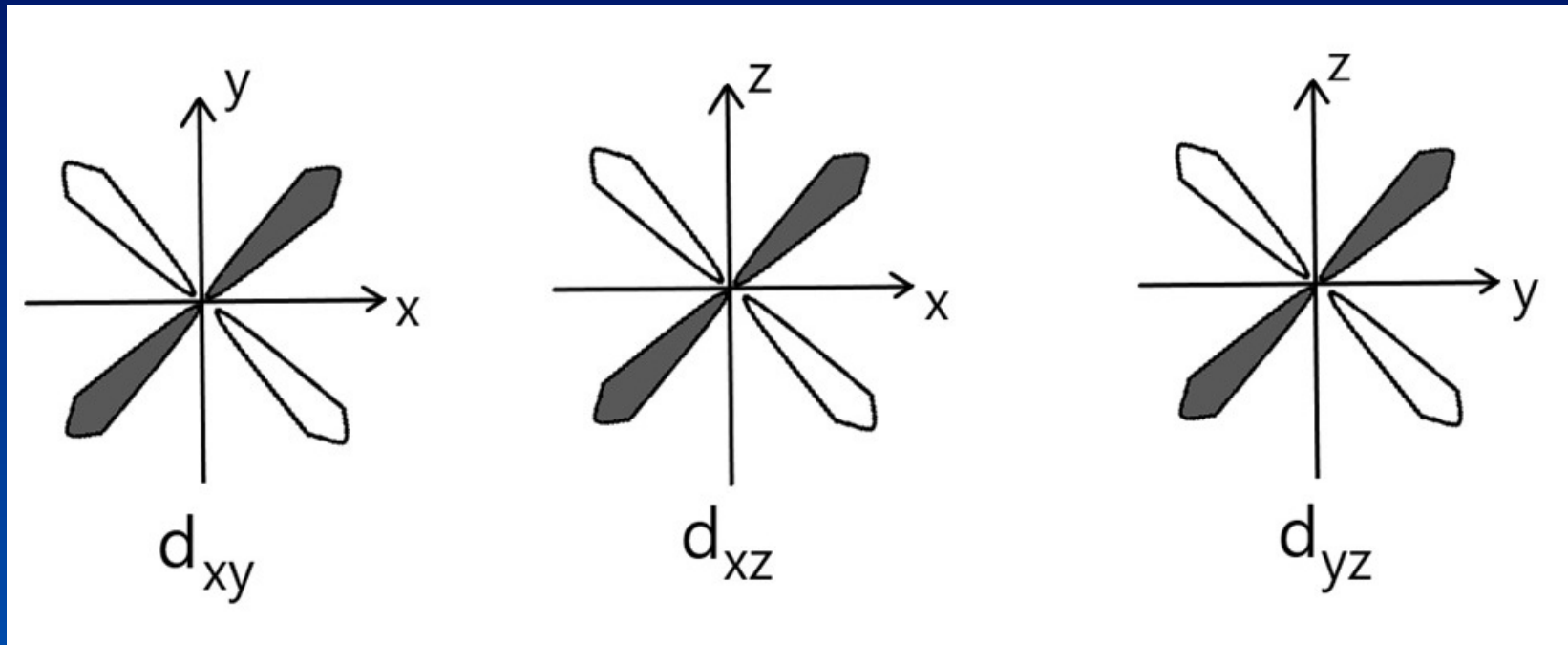
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Os orbitais  $3d_{z^2}$  e  $3d_{x^2-y^2}$  do Co



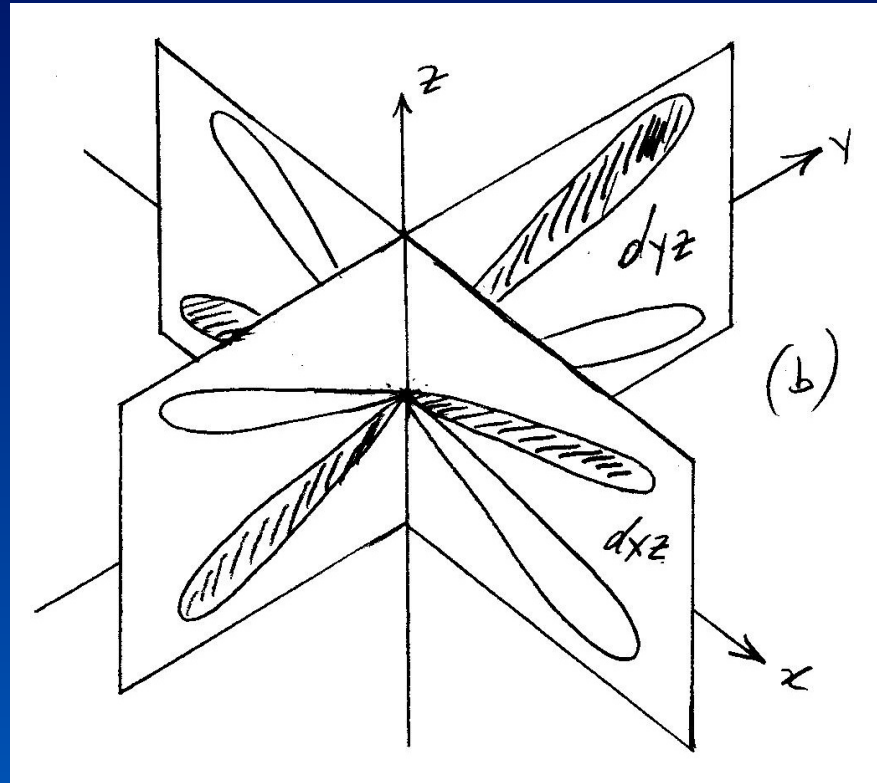
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Os orbitais  $3d_{xy}$ ,  $3d_{xz}$  e  $3d_{yz}$  do Co



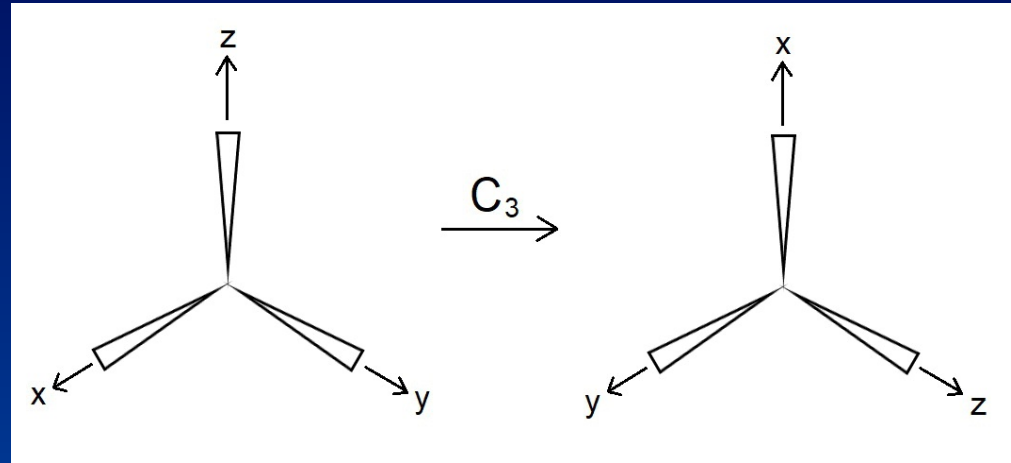


Os orbitais  $3d_{xz}$  e  $3d_{yz}$  do Co





# Transformação dos eixos na operação $C_3$



$$\begin{aligned}x &-C_3 \rightarrow y \\y &-C_3 \rightarrow z \\z &-C_3 \rightarrow x\end{aligned}$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{xy}$ ,  $3d_{xz}$ ,  $3d_{yz}$  do Co -INSEPARÁVEIS

$$x - C_3 \rightarrow y$$

$$y - C_3 \rightarrow z$$

$$z - C_3 \rightarrow x$$

$$d_{xy} - C_3 \rightarrow d_{yz}$$

$$d_{xz} - C_3 \rightarrow d_{yx} = d_{xy}$$

$$d_{yz} - C_3 \rightarrow d_{zx} = d_{xz}$$

	$d_{xy}$	$d_{xz}$	$d_{yz}$		$d_{xy}$	$d_{xz}$	$d_{yz}$
$d_{xy}$	1	0	0	$C_3$	0	0	1
$d_{xz}$	0	1	0	$\rightarrow$	$d_{xz}$	1	0
$d_{yz}$	0	0	1		$d_{yz}$	0	1

$$\chi = 0$$

Classificando os orbitais  $d_{xy}$ ,  $d_{xz}$ ,  $d_{yz}$  do Co -INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

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$3d_{mn}$  3 0

# $[\text{CoF}_6]^{3-} - \text{O}_h$

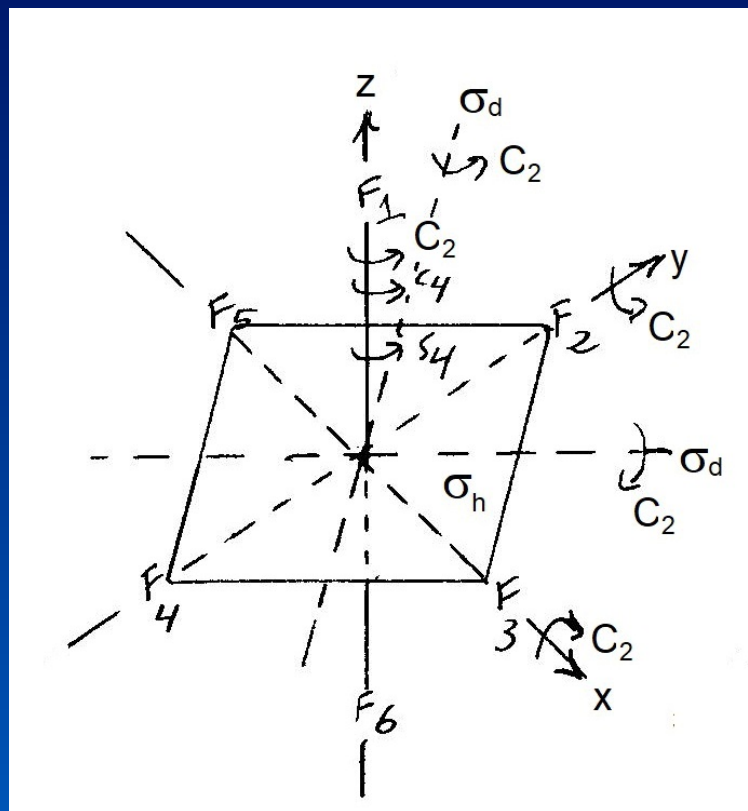
Classificando os orbitais  $3d_{xy}$ ,  $3d_{xz}$ ,  $3d_{yz}$  do Co -INSEPARÁVEIS

$C_2 \neq x, y, z$

$x - C_2 \rightarrow y$

$y - C_2 \rightarrow x$

$z - C_2 \rightarrow -z$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{xy}$ ,  $3d_{xz}$ ,  $3d_{yz}$  do Co -INSEPARÁVEIS

$$\begin{array}{l} x \quad C_2 \rightarrow y \\ y \quad C_2 \rightarrow x \\ z \quad C_2 \rightarrow -z \end{array}$$

$$\begin{array}{l} d_{xy} \quad C_2 \rightarrow d_{yx} = d_{xy} \\ d_{xz} \quad C_2 \rightarrow d_{y(-z)} = -d_{yz} \\ d_{yz} \quad C_2 \rightarrow d_{x(-z)} = -d_{xz} \end{array}$$

$$\begin{array}{ccc} & d_{xy} & d_{xz} & d_{yz} \\ d_{xy} & 1 & 0 & 0 \\ d_{xz} & 0 & 1 & 0 \\ d_{yz} & 0 & 0 & 1 \end{array} \quad C_2 \rightarrow \quad \begin{array}{ccc} & d_{xy} & d_{xz} & d_{yz} \\ d_{xy} & 1 & 0 & 0 \\ d_{xz} & 0 & 0 & -1 \\ d_{yz} & 0 & -1 & 0 \end{array}$$

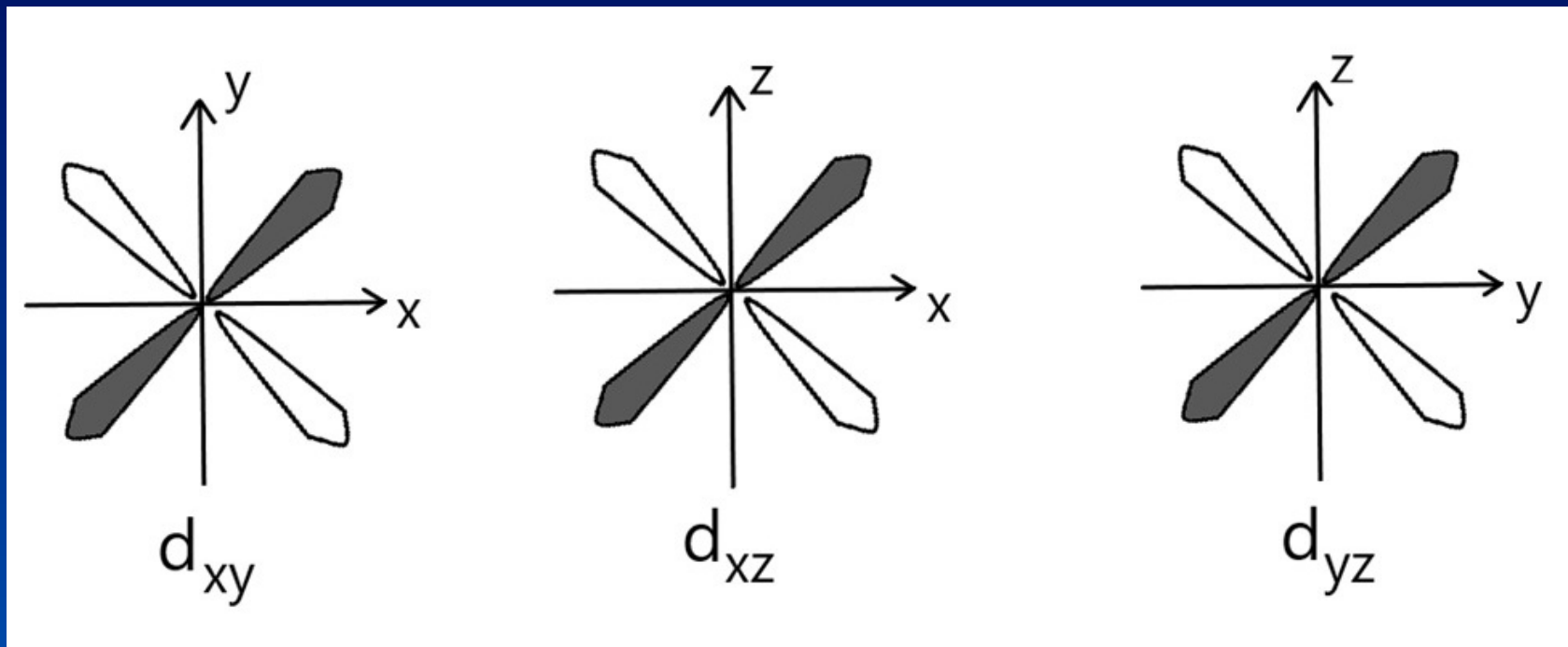
$$\chi = 1$$

Classificando os orbitais  $d_{xy}$ ,  $d_{xz}$ ,  $d_{yz}$  do Co -INSEPARÁVEIS

		(x,y,z)									
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	
$A_{1g}$	1	1	1	1	1	1	1	1	1	1	
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1	
$E_g$	2	-1	0	0	2	2	0	-1	2	0	
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1	
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1	
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1	
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1	
$E_u$	2	-1	0	0	2	-2	0	1	-2	0	
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1	
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1	
$3d_{mn}$	3	0	1								

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{xy}$ ,  $3d_{xz}$ ,  $3d_{yz}$  do Co -INSEPARÁVEIS



$i$  = centro de inversão

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{xy}$ ,  $3d_{xz}$ ,  $3d_{yz}$  do Co -INSEPARÁVEIS

$$\begin{array}{ccc} & d_{xy} & d_{xz} & d_{yz} \\ d_{xy} & 1 & 0 & 0 \\ d_{xz} & 0 & 1 & 0 \\ d_{yz} & 0 & 0 & 1 \end{array} \quad \begin{array}{c} i \\ \rightarrow \end{array} \quad \begin{array}{ccc} & d_{xy} & d_{xz} & d_{yz} \\ d_{xy} & 1 & 0 & 0 \\ d_{xz} & 0 & 1 & 0 \\ d_{yz} & 0 & 0 & 1 \end{array}$$

$$\chi = 3$$

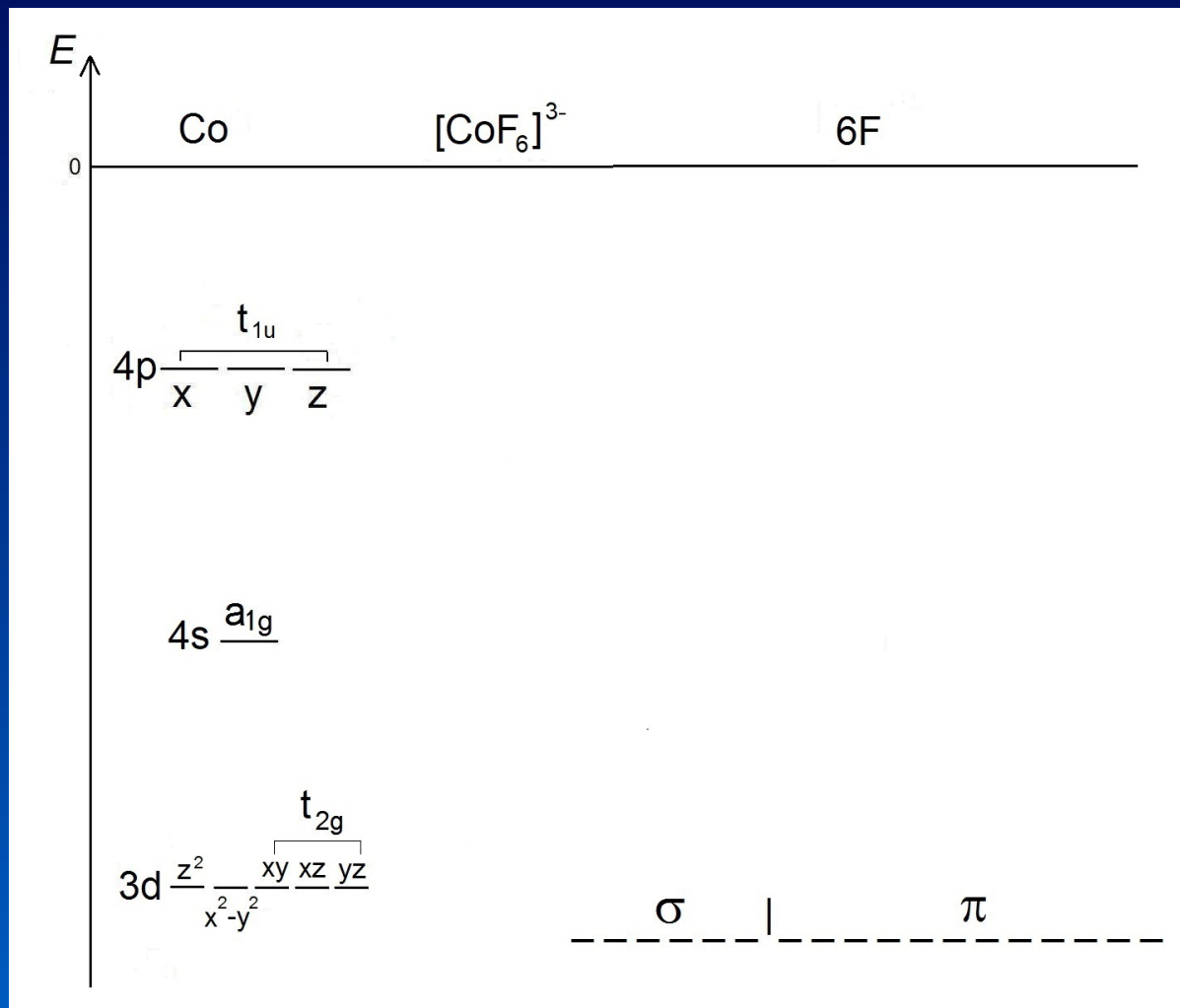


Classificando os orbitais  $d_{xy}$ ,  $d_{xz}$ ,  $d_{yz}$  do Co -INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1
$3d_{mn}$	3	0	1			3				$T_{2g}$

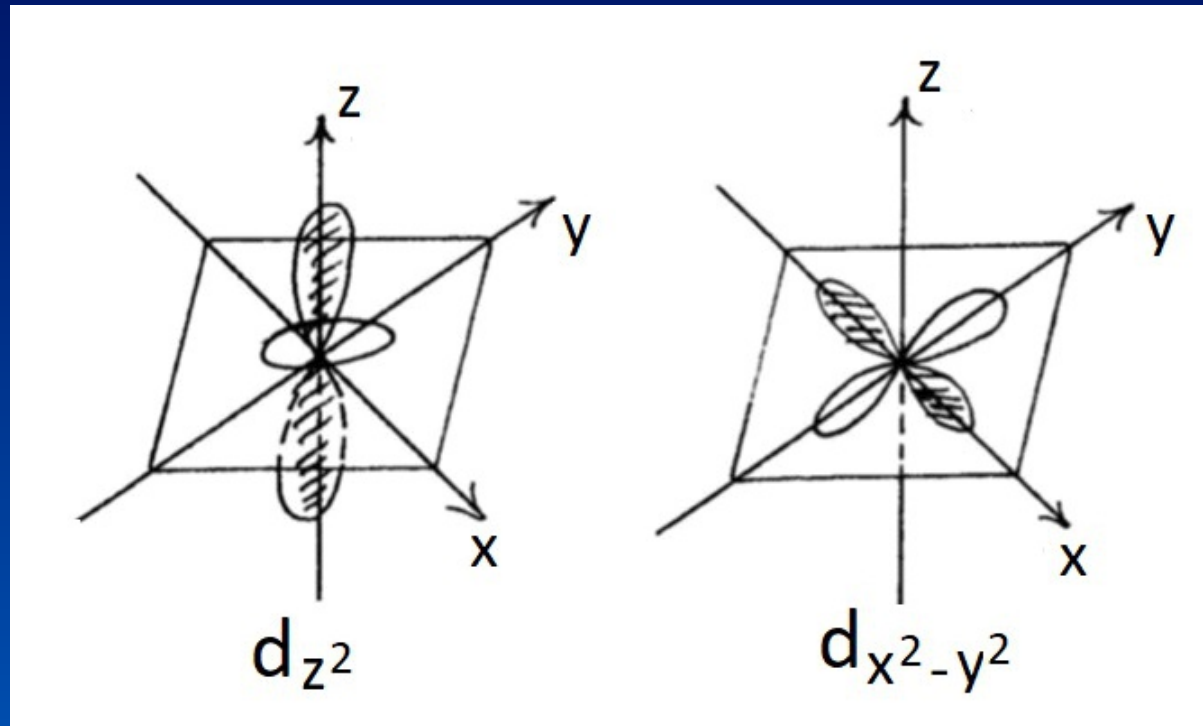
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Construindo o diagrama de energia dos orbitais moleculares



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co, INSEPARÁVEIS(!)



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co, INSEPARÁVEIS(!)

$$x - C_3 \rightarrow y$$

$$y - C_3 \rightarrow z$$

$$z - C_3 \rightarrow x$$

$$d_{z^2} \quad C_3 \rightarrow d_{x^2}$$

$$d_{x^2-y^2} \quad C_3 \rightarrow d_{y^2-z^2}$$

Na verdade, a expressão do orbital  $d_{z^2}$  é mais complexa:

$$d_{z^2} = (1/\sqrt{3})(2z^2 - x^2 - y^2) \quad C_3 \rightarrow d_{x^2} = (1/\sqrt{3})(2x^2 - y^2 - z^2)$$

$$d_{x^2-y^2} = x^2 - y^2 \quad C_3 \rightarrow d_{y^2-z^2} = y^2 - z^2$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

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Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

Pode-se mostrar que os novos orbitais  $d_{x^2}$  e  $d_{y^2-z^2}$  são uma combinação linear dos orbitais  $d_{z^2}$  e  $d_{x^2-y^2}$ .

$$d_{x^2} = (1/\sqrt{3})(2x^2 - y^2 - z^2) = C_{11} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{12}(x^2 - y^2)$$

$$d_{y^2-z^2} = y^2 - z^2 = C_{21} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{22}(x^2 - y^2)$$

Colecionando-se os termos de  $z^2$  da primeira equação, obtém-se o valor de  $C_{11}$ :

$$-(1/\sqrt{3}) = C_{11} (1/\sqrt{3}) \times 2$$

$$C_{11} = -1/2$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

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Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

$$d_{x^2} = (1/\sqrt{3})(2x^2 - y^2 - z^2) = C_{11} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{12}(x^2 - y^2)$$

$$d_{y^2-z^2} = y^2 - z^2 = C_{21} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{22}(x^2 - y^2)$$

Colecionando-se os termos de  $y^2$  da primeira equação, obtém-se o valor de  $C_{12}$ :

$$-(1/\sqrt{3}) = -C_{11} (1/\sqrt{3}) - C_{12}$$

$$-(1/\sqrt{3}) = -(-1/2)(1/\sqrt{3}) - C_{12}$$

$$C_{12} = -(1/\sqrt{3}) - (1/2)(1/\sqrt{3}) = (\sqrt{3})/2$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

---

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

$$d_{x^2} = (1/\sqrt{3})(2x^2 - y^2 - z^2) = C_{11} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{12}(x^2 - y^2)$$

$$d_{y^2-z^2} = y^2 - z^2 = C_{21} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{22}(x^2 - y^2)$$

Colecionando-se os termos de  $z^2$  da segunda equação, obtém-se o valor de  $C_{21}$ :

$$-1 = C_{21} (2/\sqrt{3})$$

$$C_{21} = -(\sqrt{3})/2$$

# $[\text{CoF}_6]^{3-} - \text{O}_h$

---

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

$$d_{x^2} = (1/\sqrt{3})(2x^2 - y^2 - z^2) = C_{11} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{12}(x^2 - y^2)$$

$$d_{y^2-z^2} = y^2 - z^2 = C_{21} (1/\sqrt{3})(2z^2 - x^2 - y^2) + C_{22}(x^2 - y^2)$$

Colecionando-se os termos de  $y^2$  da segunda equação, obtém-se o valor de  $C_{22}$ :

$$1 = -C_{21} (1/\sqrt{3}) - C_{22}$$

$$1 = (\sqrt{3}/2)(1/\sqrt{3}) - C_{22}$$

$$C_{22} = -1 + 1/2 = \mathbf{-1/2}$$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

$$\begin{array}{cc} & \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} \\ \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} & \begin{array}{cc} d_{z^2} & d_{x^2-y^2} \\ 1 & 0 \\ 0 & 1 \end{array} & \begin{array}{c} C_3 \\ \rightarrow \end{array} & \begin{array}{cc} d_{z^2} & d_{x^2-y^2} \\ -1/2 & -(\sqrt{3})/2 \\ -(\sqrt{3})/2 & -1/2 \end{array} \end{array}$$

$$\chi = -1$$

Classificando os orbitais  $d_{z^2}$ ,  $d_{x^2-y^2}$  do Co -INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

---

$d_{z^2, x^2-y^2}$  2 -1

# $[\text{CoF}_6]^{3-} - \text{O}_h$

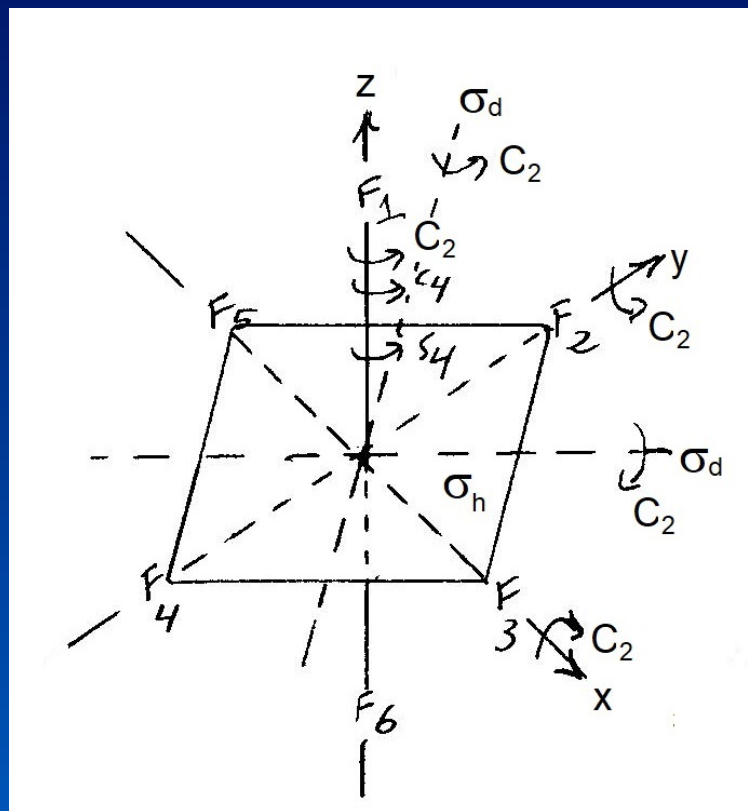
Classificando os orbitais  $3d_{z^2}$  e  $3d_{x^2-y^2}$ , do Co, INSEPARÁVEIS

$C_2 \neq x, y, z$

x  $C_2 \rightarrow y$

y  $C_2 \rightarrow x$

z  $C_2 \rightarrow -z$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co, INSEPARÁVEIS

$$\begin{array}{l}
 \text{x} \quad C_2 \rightarrow \text{y} \\
 \text{y} \quad C_2 \rightarrow \text{x} \\
 \text{z} \quad C_2 \rightarrow -\text{z}
 \end{array}
 \quad
 \begin{array}{l}
 d_{z^2} \quad C_2 \rightarrow d_{(-z)(-z)} = d_{z^2} \\
 d_{x^2-y^2} \quad C_2 \rightarrow d_{y^2-x^2} = -d_{x^2-y^2}
 \end{array}$$

$$\begin{array}{ccc}
 & d_{z^2} & d_{x^2-y^2} \\
 d_{z^2} & 1 & 0 \\
 d_{x^2-y^2} & 0 & 1
 \end{array}
 \xrightarrow{C_2}
 \begin{array}{ccc}
 & d_{z^2} & d_{x^2-y^2} \\
 d_{z^2} & 1 & 0 \\
 d_{x^2-y^2} & 0 & -1
 \end{array}$$

$\chi = 2$ 
 $\chi = 0$

Classificando os orbitais  $d_{z^2}$ ,  $d_{x^2-y^2}$  do Co -INSEPARÁVEIS

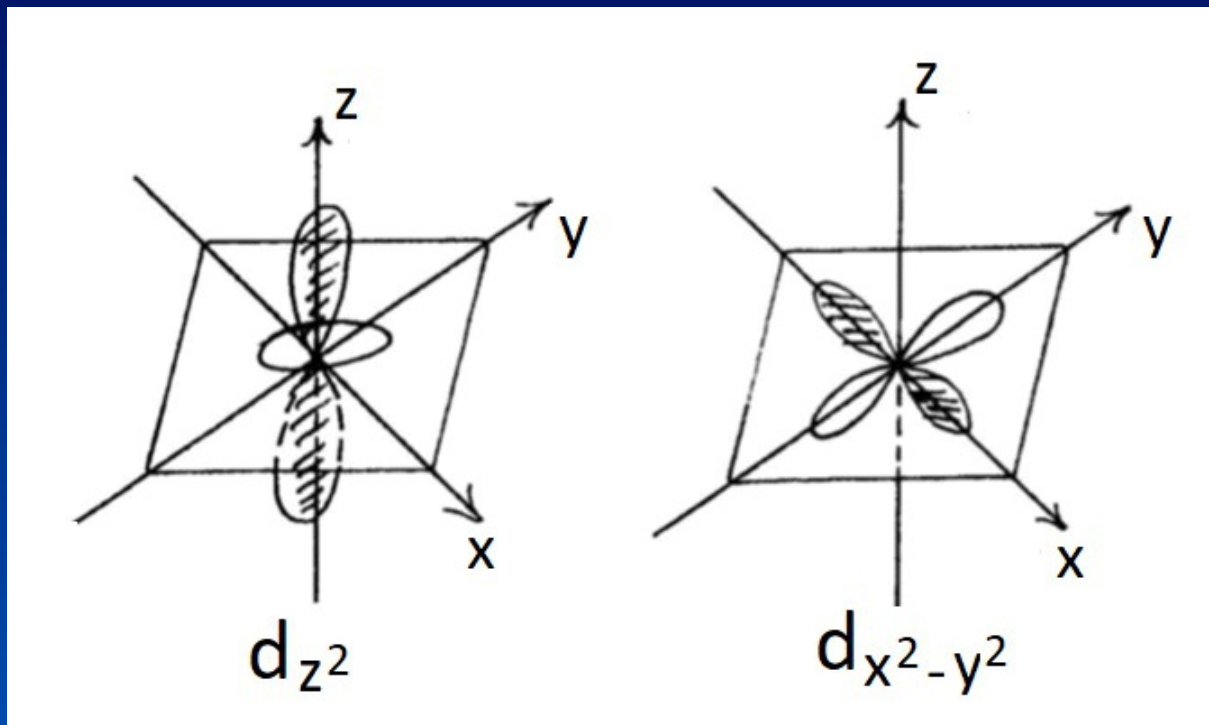
					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

---

$d_{z^2}, d_{x^2-y^2}$     2    -1    0

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS



i = centro de inversão

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $3d_{z^2}$ ,  $3d_{x^2-y^2}$  do Co -INSEPARÁVEIS

$$\begin{array}{cc} & \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} \\ \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} & \begin{array}{cc} 1 & 0 \\ 0 & 1 \end{array} \end{array} \quad \begin{array}{c} i \\ \rightarrow \end{array} \quad \begin{array}{cc} & \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} \\ \begin{array}{c} d_{z^2} \\ d_{x^2-y^2} \end{array} & \begin{array}{cc} 1 & 0 \\ 0 & 1 \end{array} \end{array}$$

$\chi = 2$

Classificando os orbitais  $d_{z^2}$ ,  $d_{x^2-y^2}$  do Co -INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

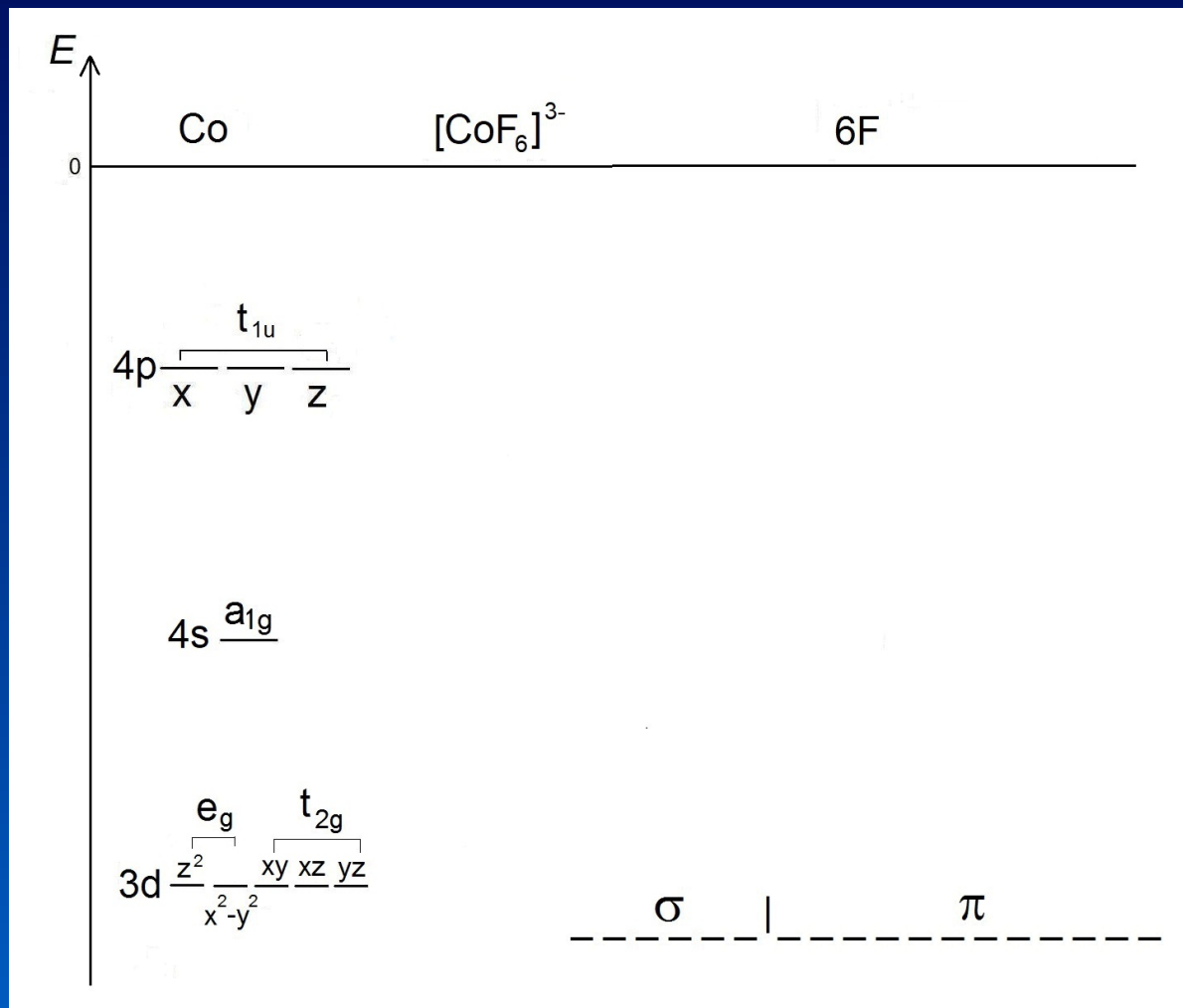
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$z^2, x^2-y^2$	2	-1	0			2				$E_g$
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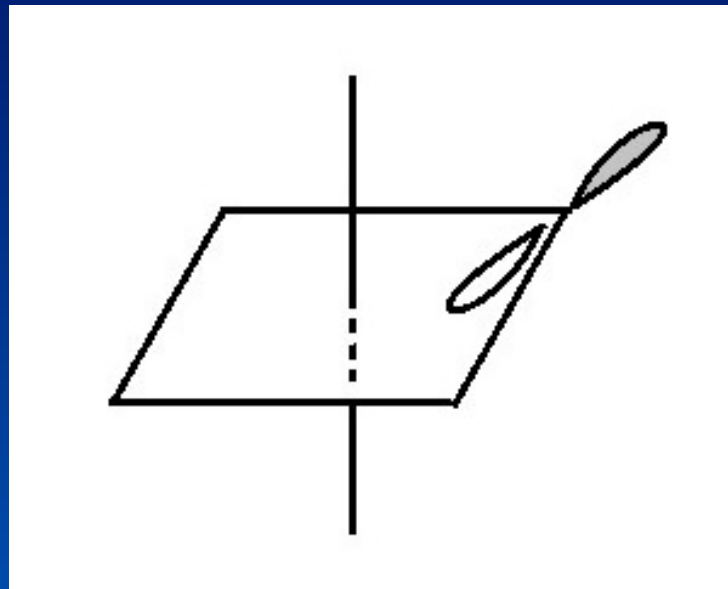
# $[\text{CoF}_6]^{3-} - \text{O}_h$

Construindo o diagrama de energia dos orbitais moleculares



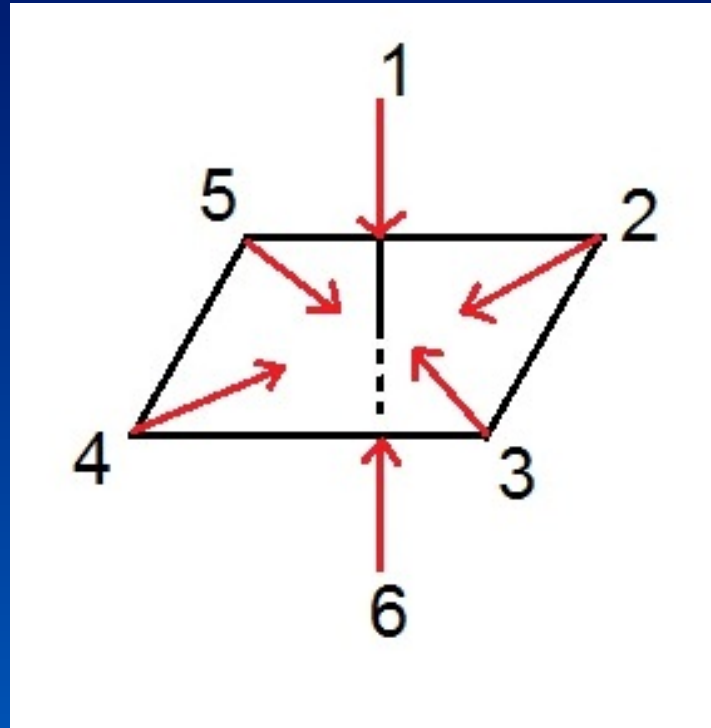


Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS

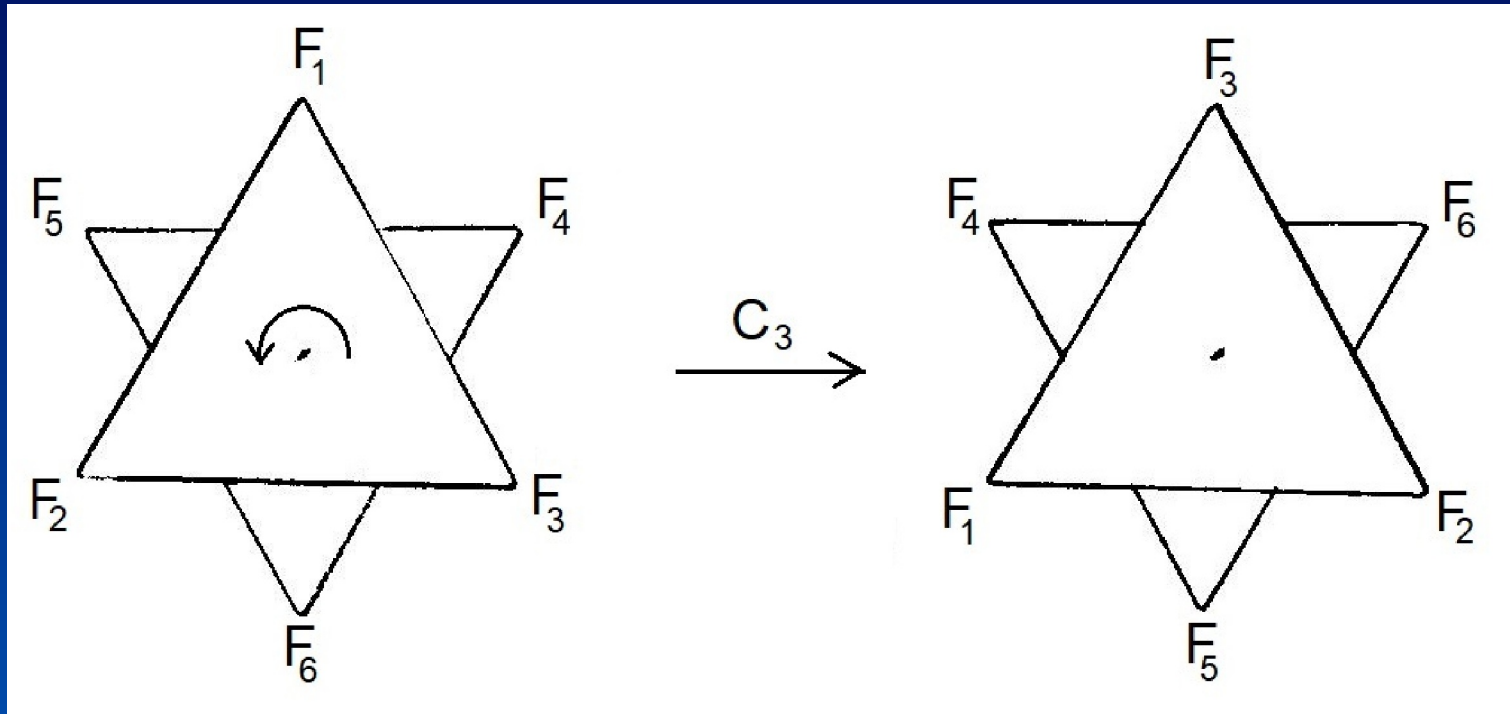




Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



# $[\text{CoF}_6]^{3-} - \text{O}_h$



# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS

	$\sigma_1$	$\sigma_2$	$\sigma_3$	$\sigma_4$	$\sigma_5$	$\sigma_6$
$\sigma_1$	1	0	0	0	0	0
$\sigma_2$	0	1	0	0	0	0
$\sigma_3$	0	0	1	0	0	0
$\sigma_4$	0	0	0	1	0	0
$\sigma_5$	0	0	0	0	1	0
$\sigma_6$	0	0	0	0	0	1

$$\chi = 6$$

$C_3$   
 $\rightarrow$

	$\sigma_1$	$\sigma_2$	$\sigma_3$	$\sigma_4$	$\sigma_5$	$\sigma_6$
$\sigma_1$	0	1	0	0	0	0
$\sigma_2$	0	0	1	0	0	0
$\sigma_3$	1	0	0	0	0	0
$\sigma_4$	0	0	0	0	1	0
$\sigma_5$	0	0	0	0	0	1
$\sigma_6$	1	0	0	0	0	0

$$\chi = 0$$

## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

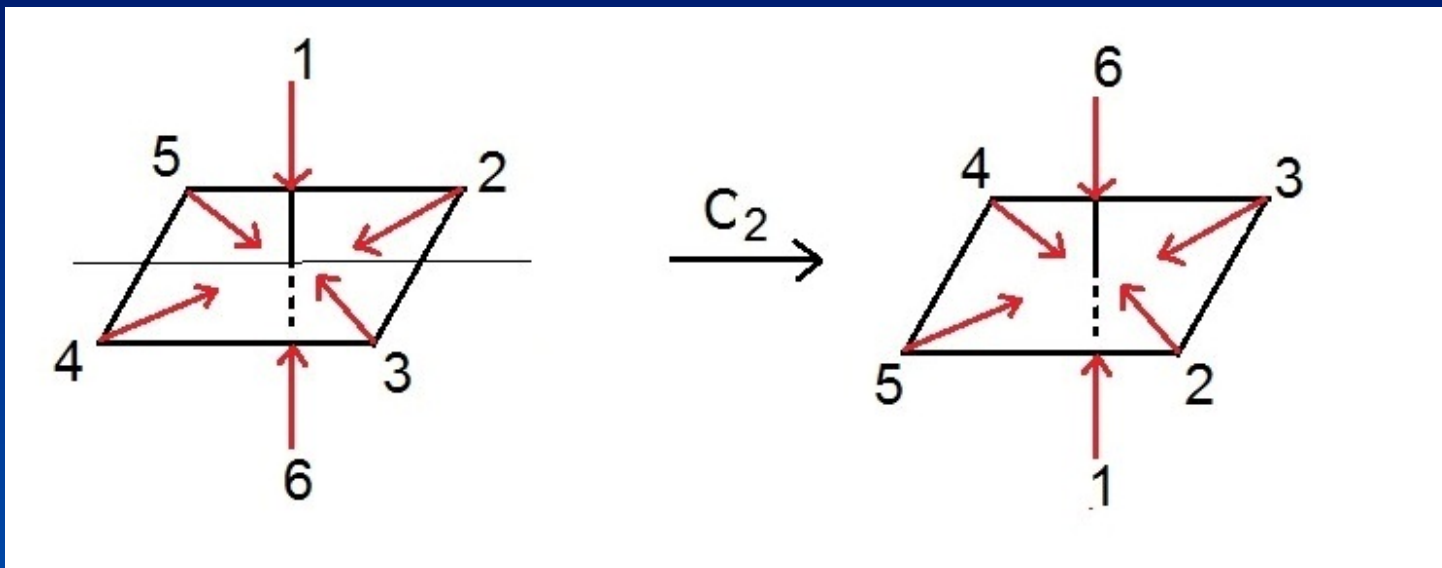
					$(x,y,z)$						
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	
$A_{1g}$	1	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	1	-1

---

$p\sigma$     6    0

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS

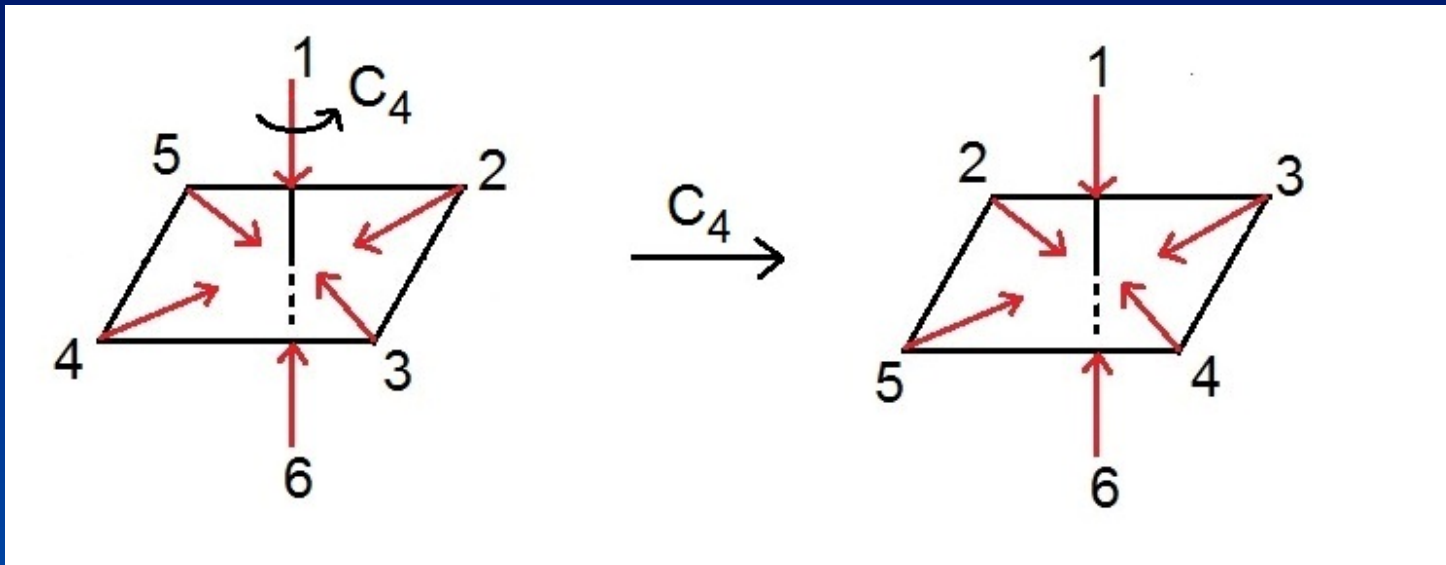






# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS

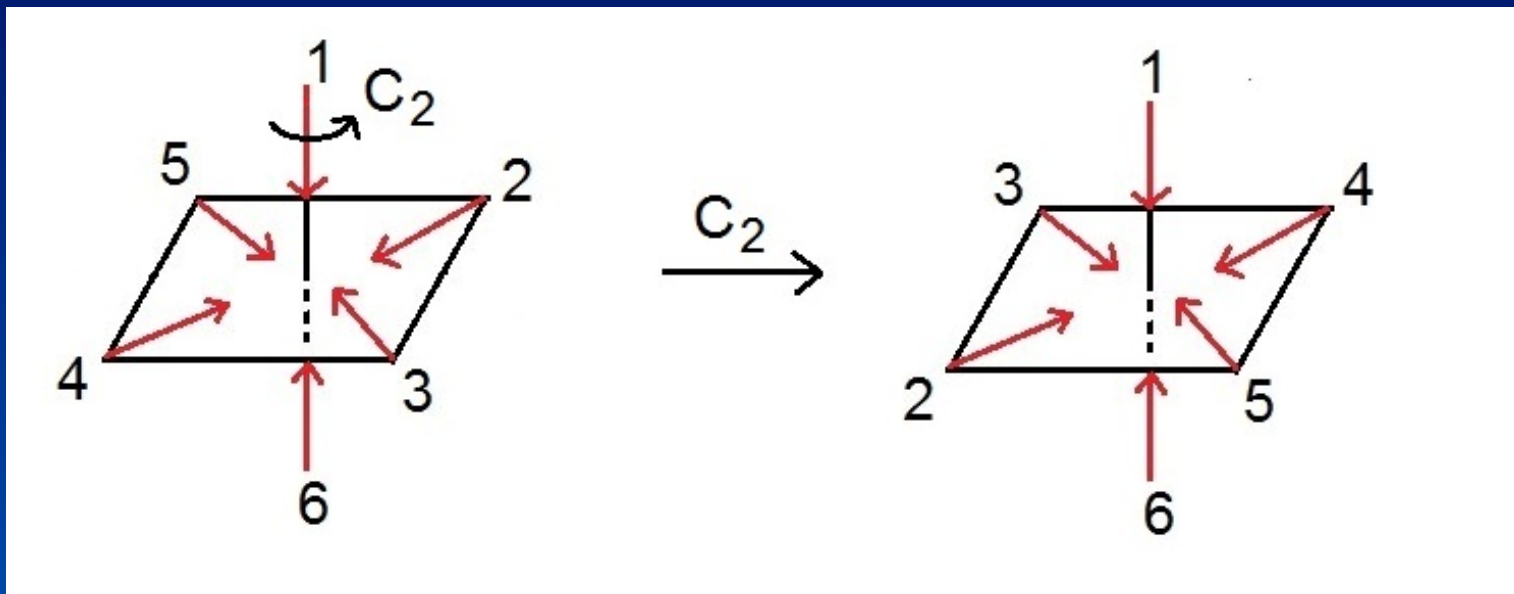


## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1
$p\sigma$	6	0	0	2						

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

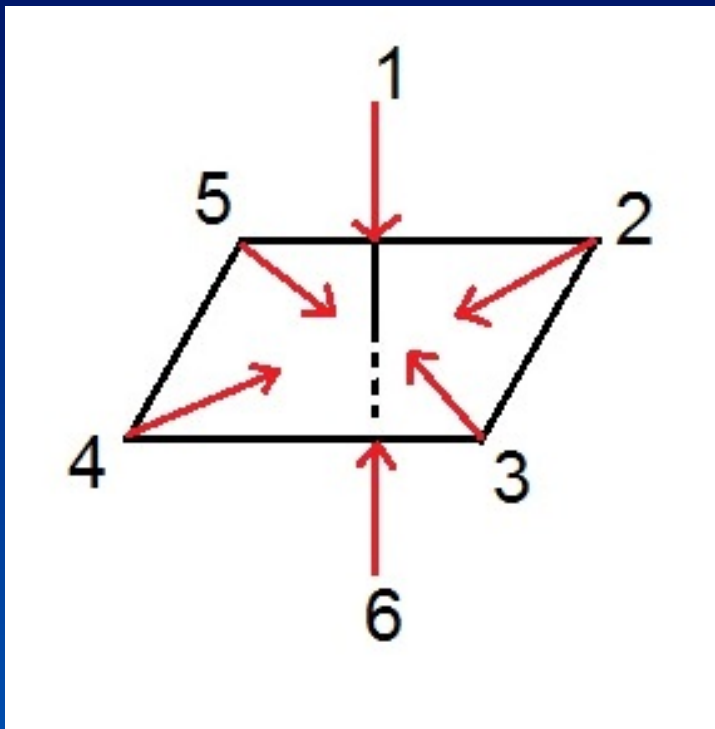
					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

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$p\sigma$	6	0	0	2	2					
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# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



$i =$  centro de inversão

todos saem do lugar  
e da diagonal da matriz  $6 \times 6$

$$\chi = 0$$

## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

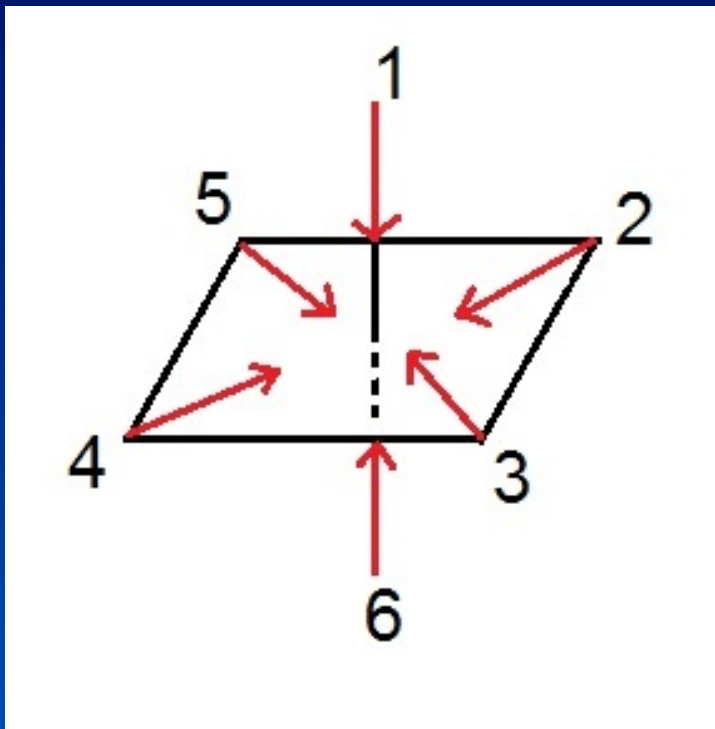
					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

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$p\sigma$	6	0	0	2	2	0				
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# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



$S_4$

todos saem do lugar  
e da diagonal da matriz  $6 \times 6$

$$\chi = 0$$

## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

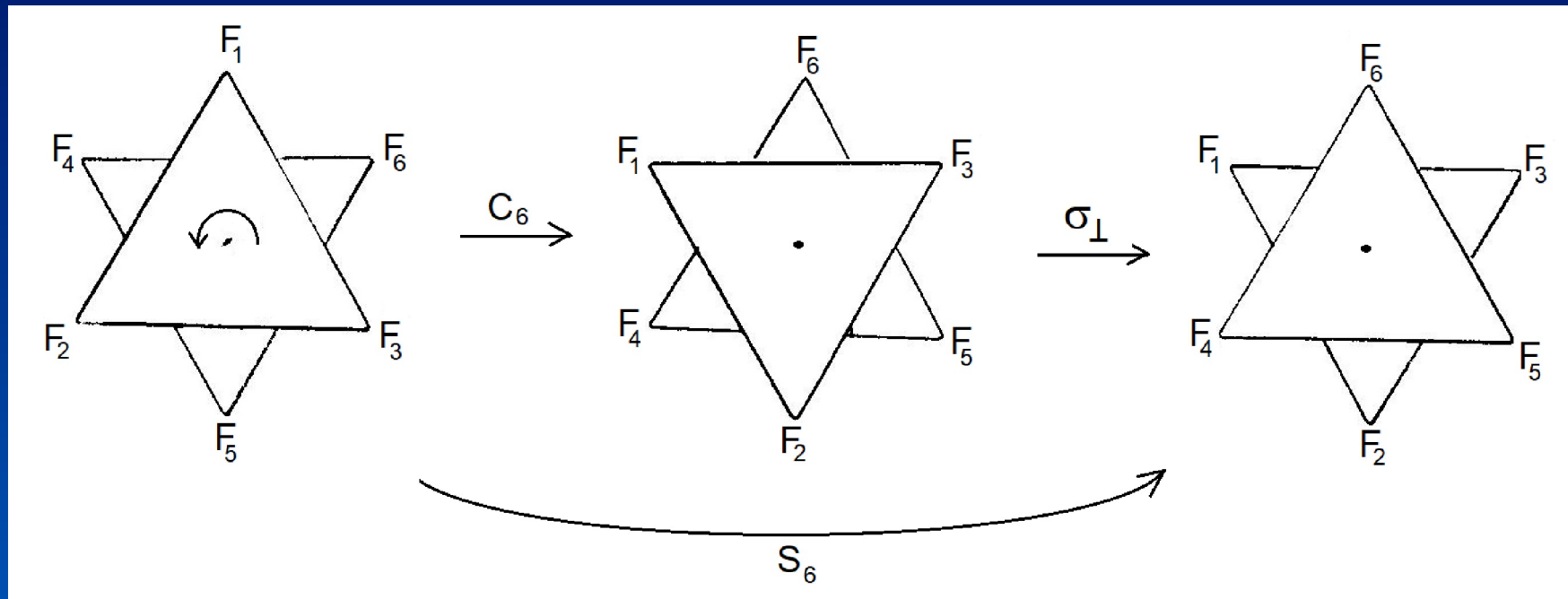
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$p\sigma$	6	0	0	2	2	0	0			
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# $[\text{CoF}_6]^{3-} - \text{O}_h$

$$S_6 = C_6 + \sigma_{\perp}$$

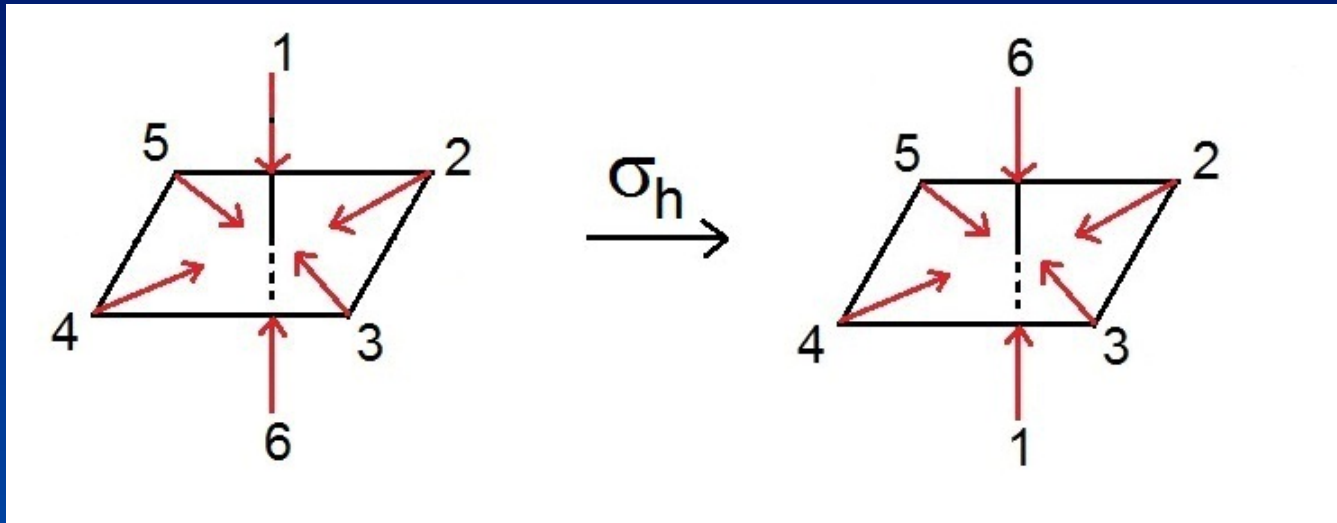


## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1
$p\sigma$	6	0	0	2	2	0	0	0		

# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

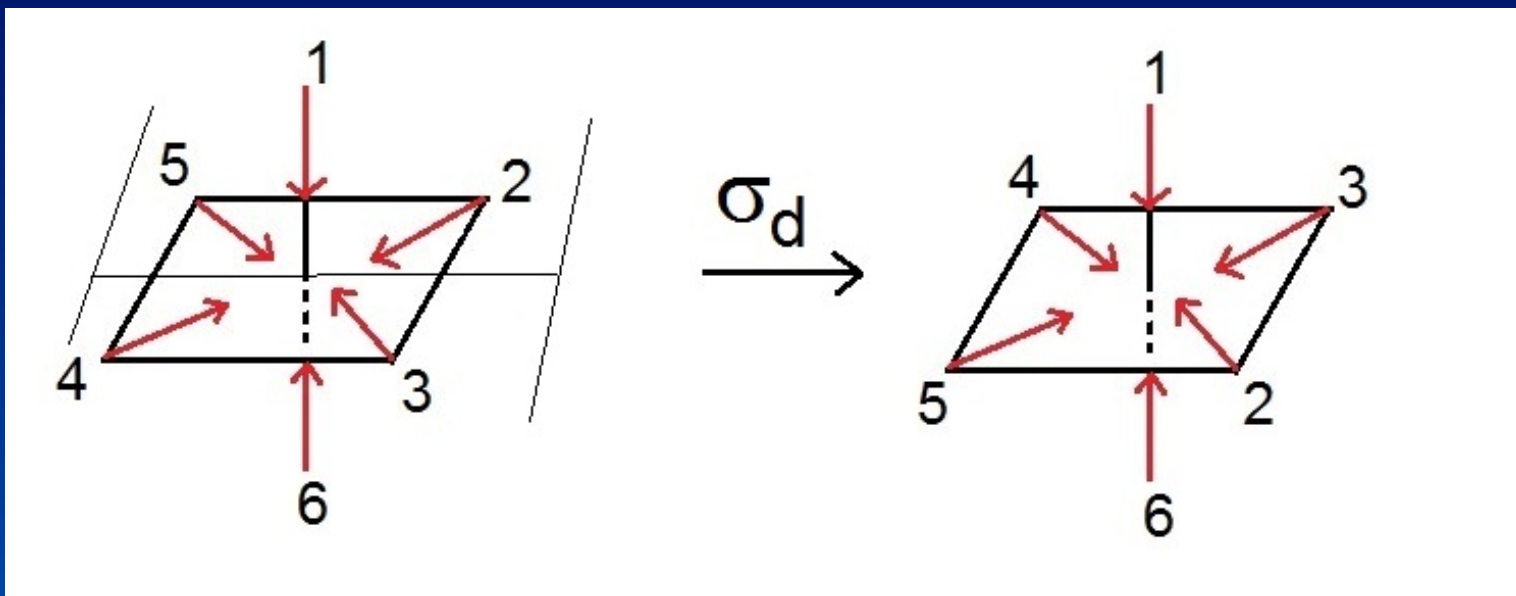
					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

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$p\sigma$	6	0	0	2	2	0	0	0	4	
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# $[\text{CoF}_6]^{3-} - \text{O}_h$

Classificando os orbitais  $2p\sigma$  dos 6F - INSEPARÁVEIS



## Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

					(x,y,z)					
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$A_{1u}$	1	1	1	1	1	-1	-1	-1	-1	-1
$A_{2u}$	1	1	-1	-1	1	-1	1	-1	-1	1
$E_u$	2	-1	0	0	2	-2	0	1	-2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1

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$p\sigma$	6	0	0	2	2	0	0	0	4	2
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# Fórmula de decomposição de Representações Redutíveis



	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	1	1	1	1	1	1	1	1	1	1
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$p\sigma$	6	0	0	2	2	0	0	0	4	2

$$48a_i = 6 + 0 + 0 + 2 \times 6 + 2 \times 3 + 0 + 0 + 0 + 4 \times 3 + 2 \times 6 = 48$$

$$a(A_{1g}) = 1$$

# Fórmula de decomposição de Representações Redutíveis



	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	1	1	-1	-1	1	1	-1	1	1	-1
$A_{2g}$	1	1	-1	-1	1	1	-1	1	1	-1
$p\sigma$	6	0	0	2	2	0	0	0	4	2

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$$48a_i = 6 + 0 + 0 - 2 \times 6 + 2 \times 3 + 0 + 0 + 0 + 4 \times 3 - 2 \times 6 = 0$$

$$a(A_{2g}) = 0$$



# Fórmula de decomposição de Representações Redutíveis



		(x,y,z)									
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	
$E_g$	2	-1	0	0	2	2	0	-1	2	0	
$p\sigma$	6	0	0	2	2	0	0	0	4	2	

---


$$48a_i = 12 + 0 + 0 + 0 + 4 \times 3 + 0 + 0 + 0 + 8 \times 3 + 0 = 48$$

$$a(E_g) = 1$$

# Fórmula de decomposição de Representações Redutíveis



		(x,y,z)									
$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2(=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1	
$p\sigma$	6	0	0	2	2	0	0	0	4	2	

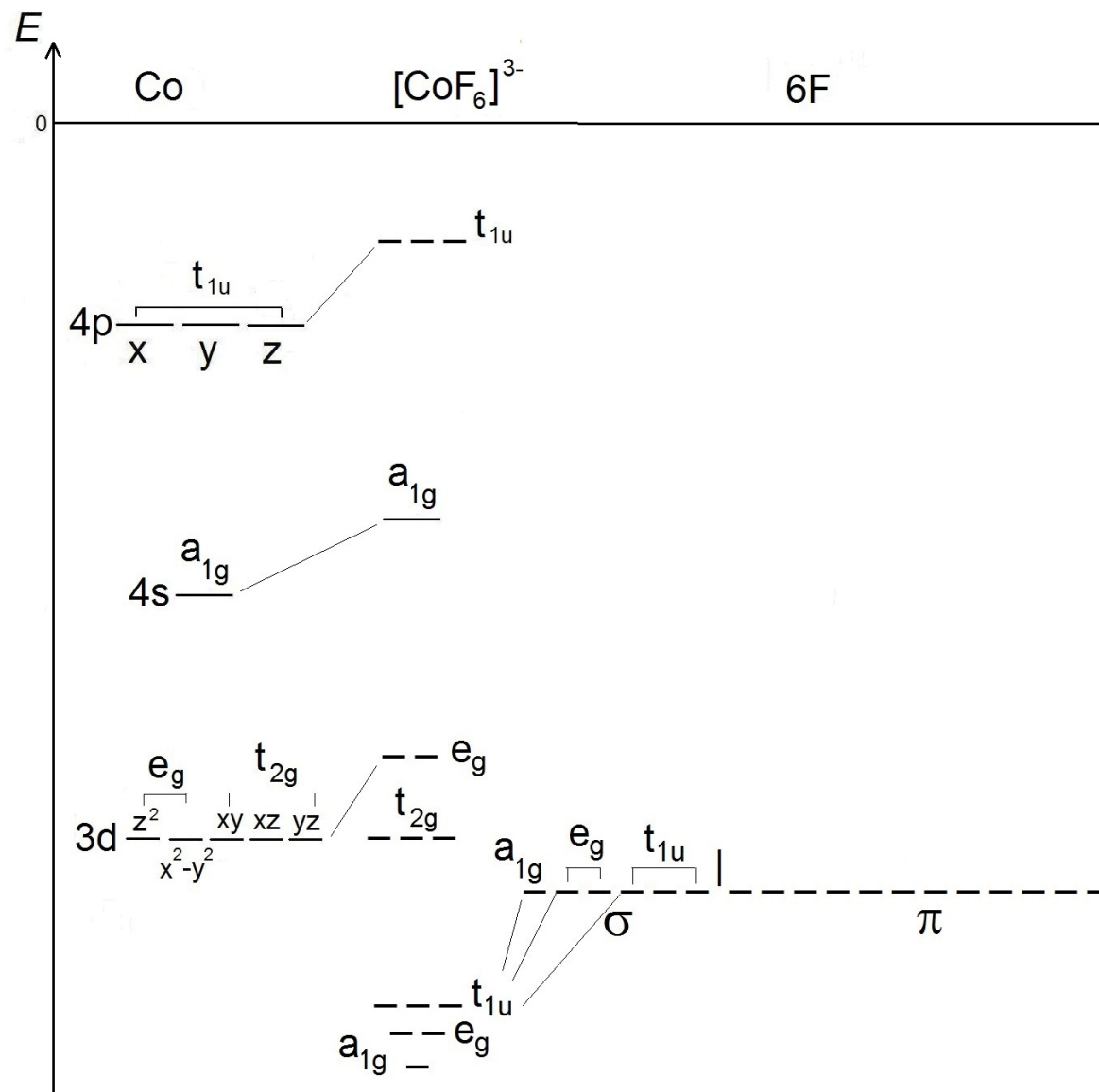
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$$48a_i = 18 + 0 + 0 + 2 \times 6 - 2 \times 3 + 0 + 0 + 0 + 4 \times 3 + 2 \times 6 = 48$$

$$a(T_{1u}) = 1$$

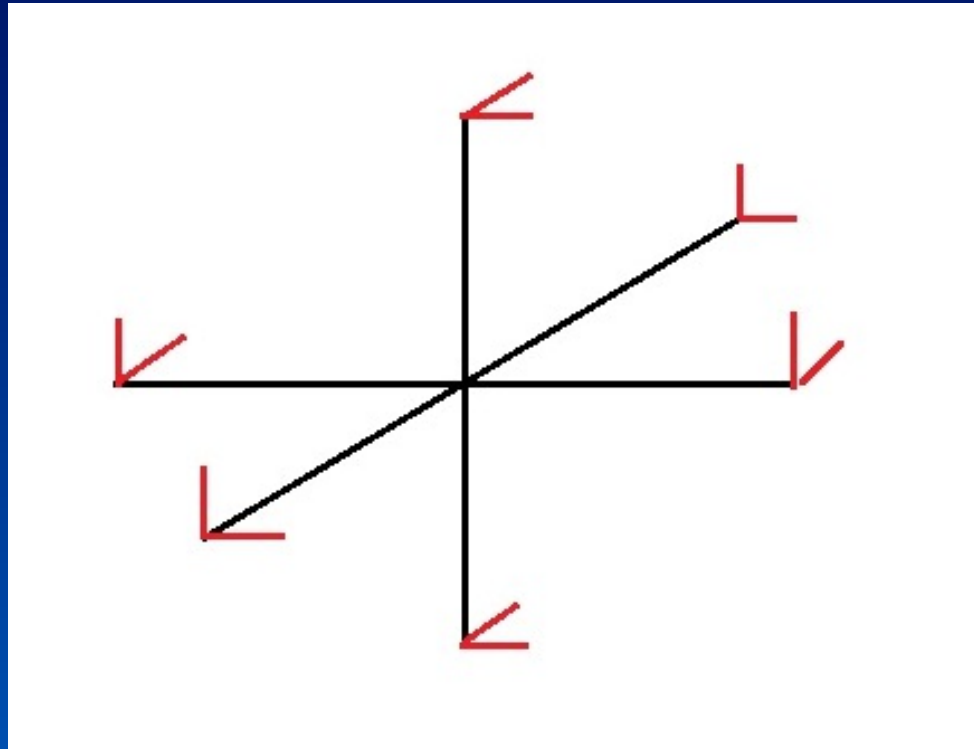
# Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2^{(x,y,z)} (=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2^{(x,y,z)} (=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$A_{1g}$	1	1	1	1	1	1	1	1	1	1
$E_g$	2	-1	0	0	2	2	0	-1	2	0
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$p\sigma$	6	0	0	2	2	0	0	0	4	2



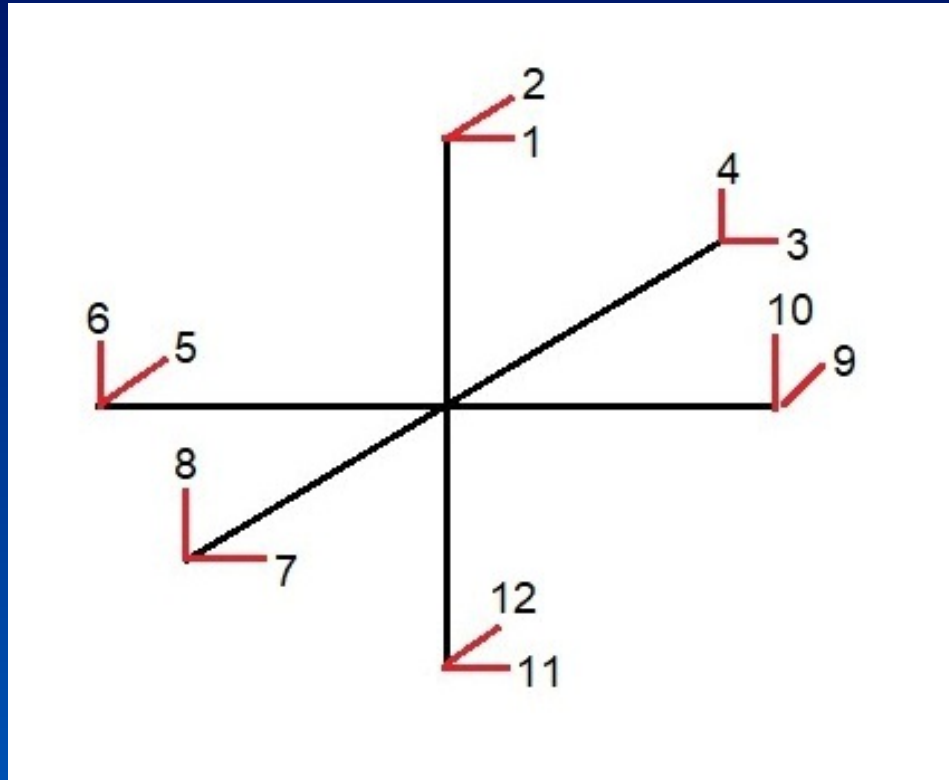


Classificando os orbitais  $2p\pi$  dos 6F - INSEPARÁVEIS





Classificando os orbitais  $2p\pi$  dos 6F - INSEPARÁVEIS





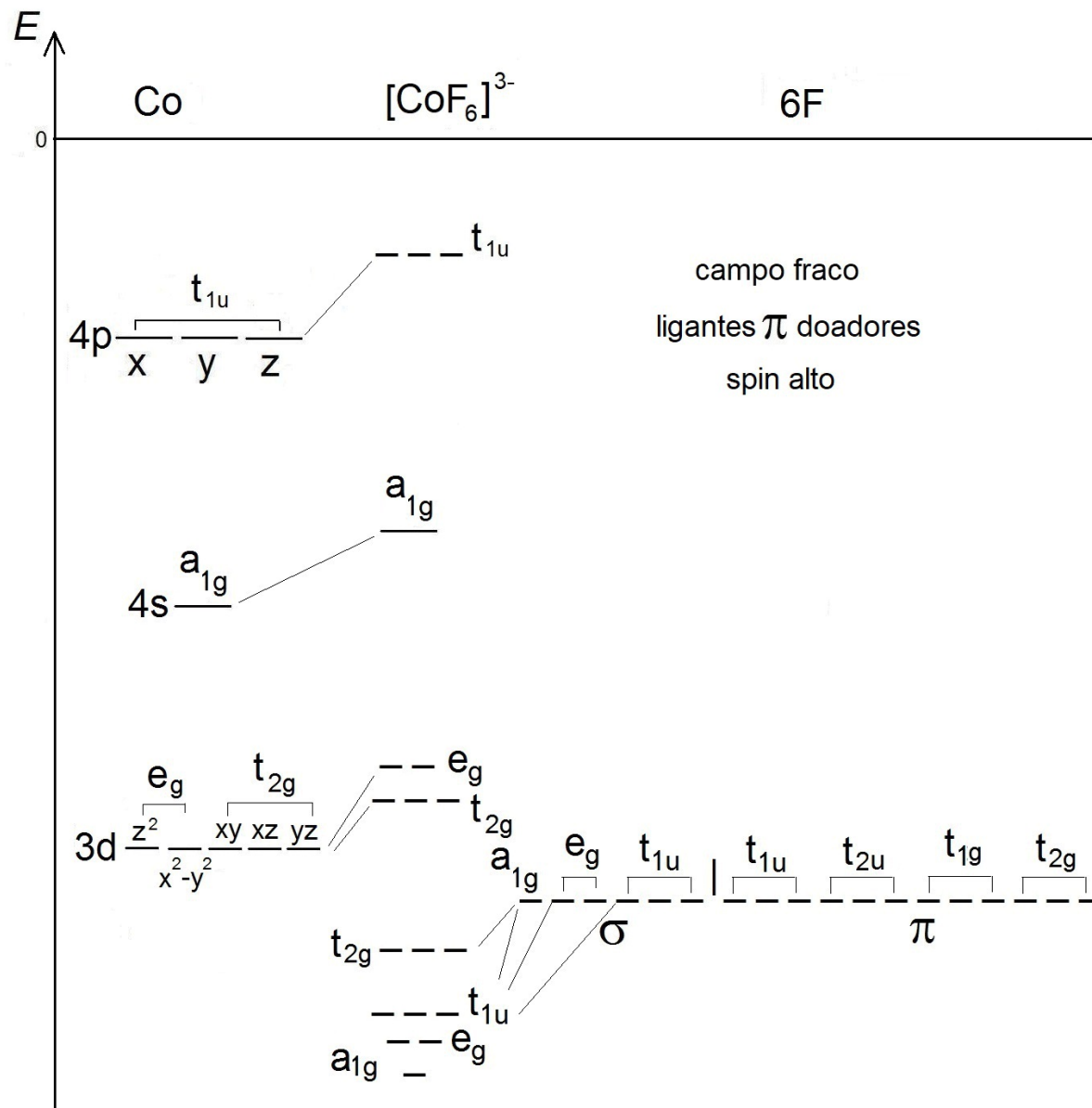
# Classificando os 12 orbitais $2p\pi$ dos $6F$ - INSEPARÁVEIS

$O_h$	E	$8C_3$	$6C_2$	$6C_4$	$3C_2^{(x,y,z)} (=C_4^2)$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$p\pi$	12	0	0	0	-4	0	0	0	0	0



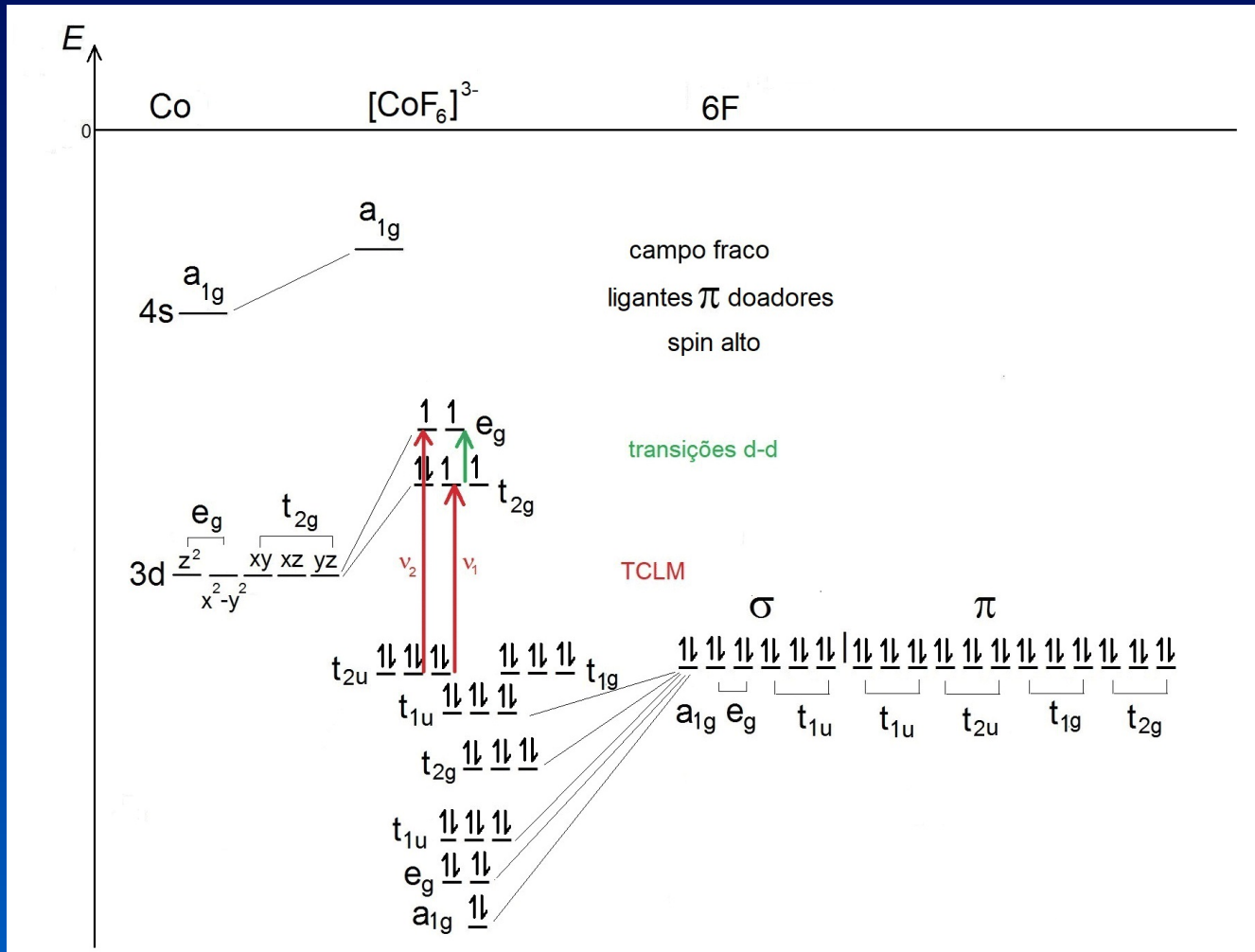
# Classificando os orbitais $2p\sigma$ dos $6F$ - INSEPARÁVEIS

	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2^{(x,y,z)} (=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2^{(x,y,z)} (=C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$T_{1g}$	3	0	-1	1	-1	3	1	0	-1	-1
$T_{2g}$	3	0	1	-1	-1	3	-1	0	-1	1
$T_{1u}$	3	0	-1	1	-1	-3	-1	0	1	1
$T_{2u}$	3	0	1	-1	-1	-3	1	0	1	-1
$\rho\pi$	12	0	0	0	-4	0	0	0	0	0

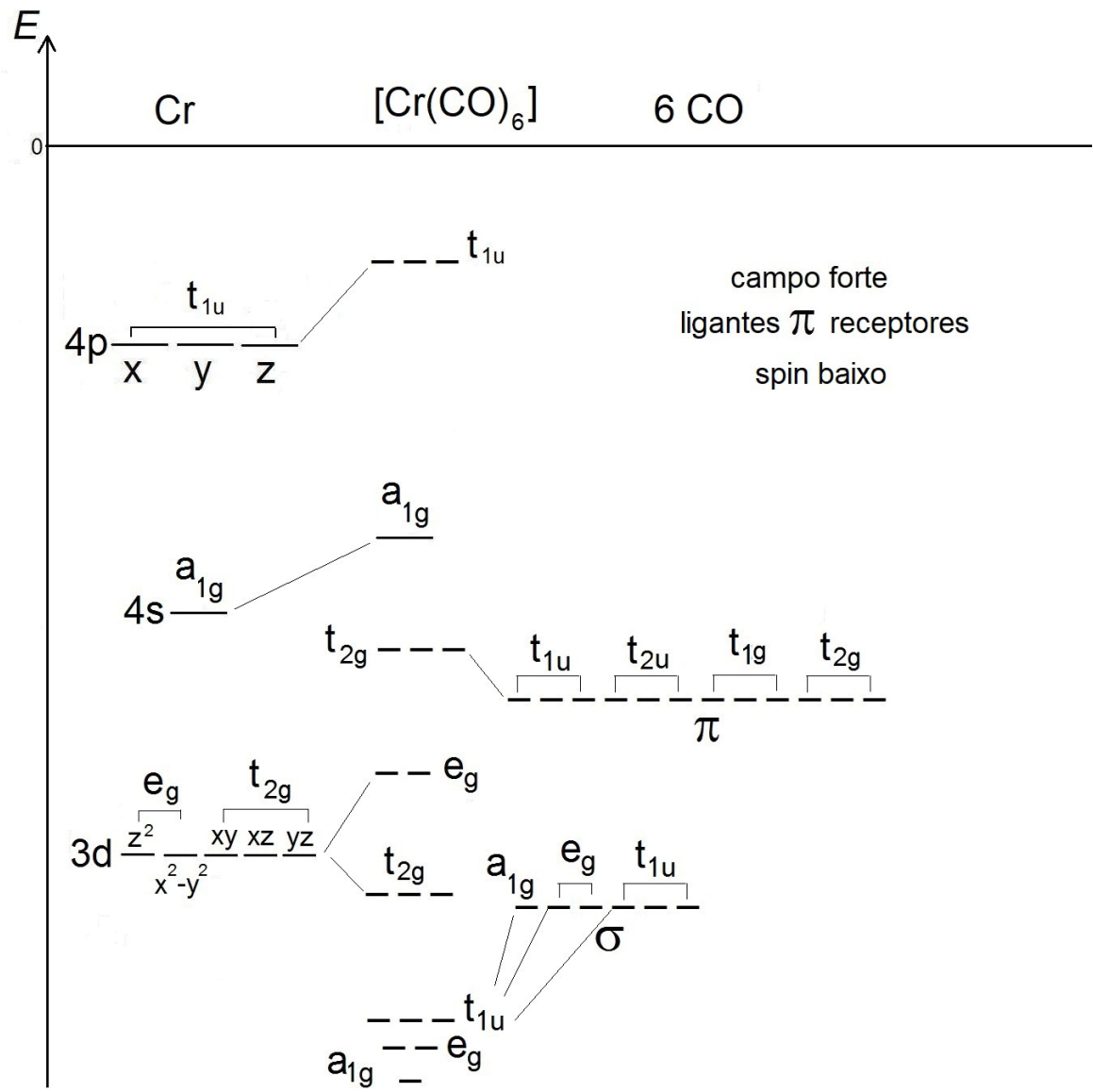


Com ligantes  $\pi$   
de campo fraco

# Transições eletrônicas e espectroscopia

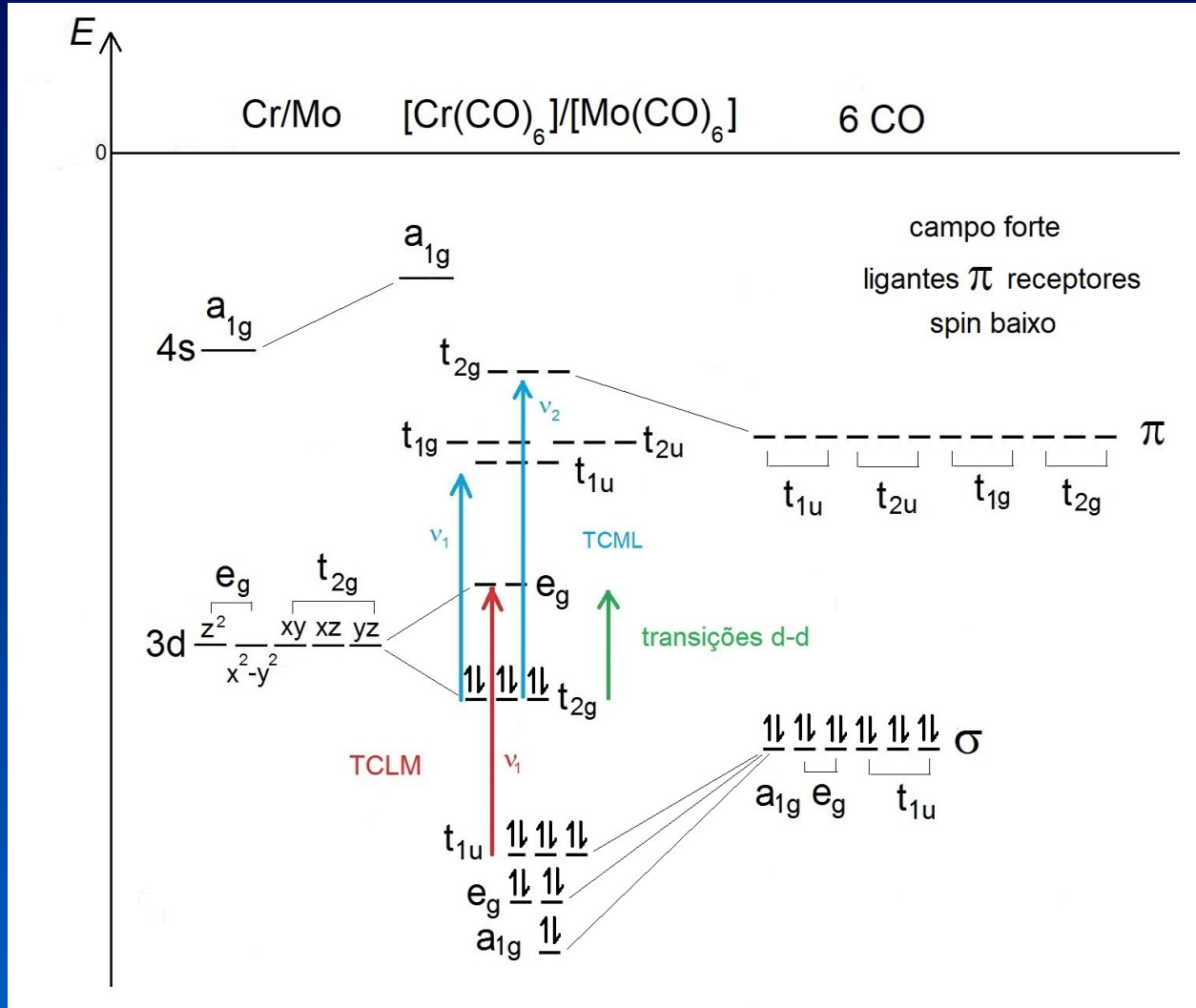


Com ligantes  $\pi$   
de campo fraco

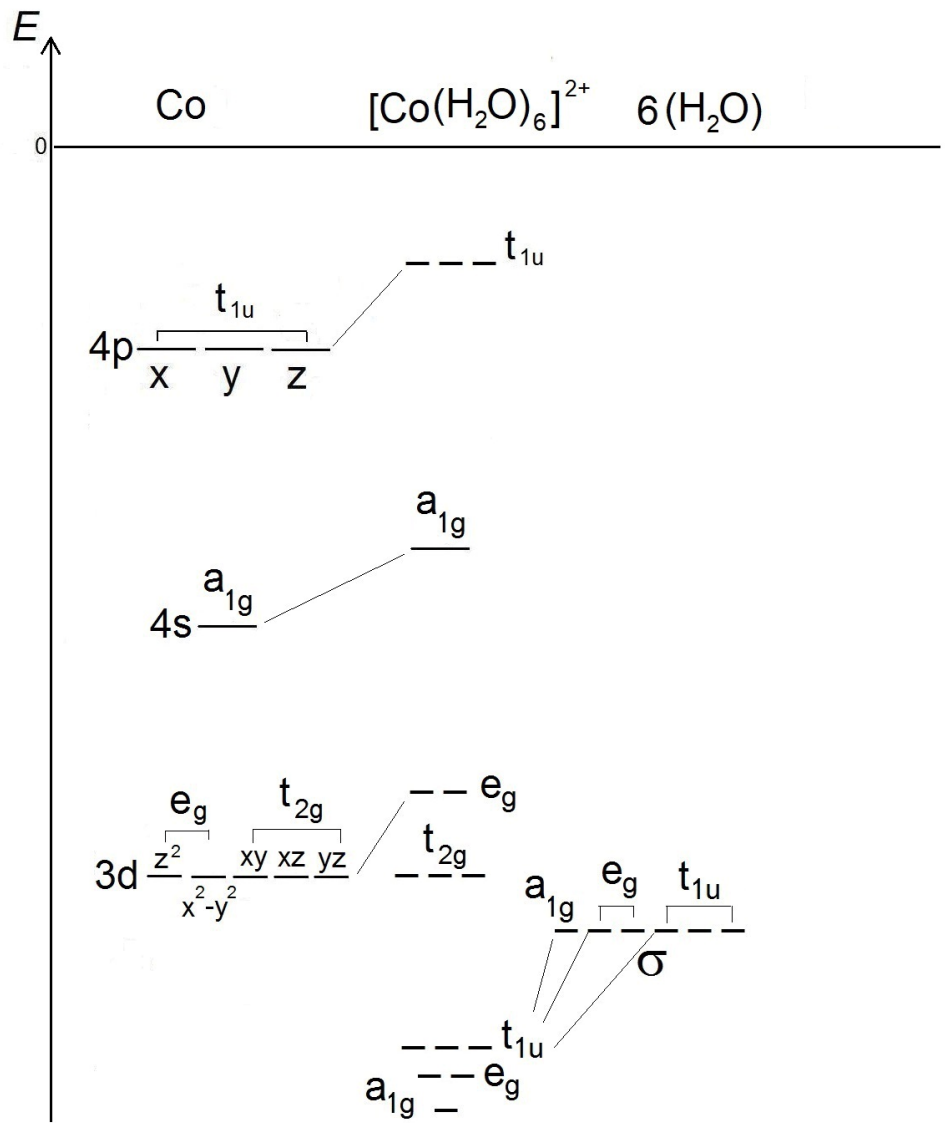


Com ligantes  $\pi$   
de campo forte

# Transições eletrônicas e espectroscopia

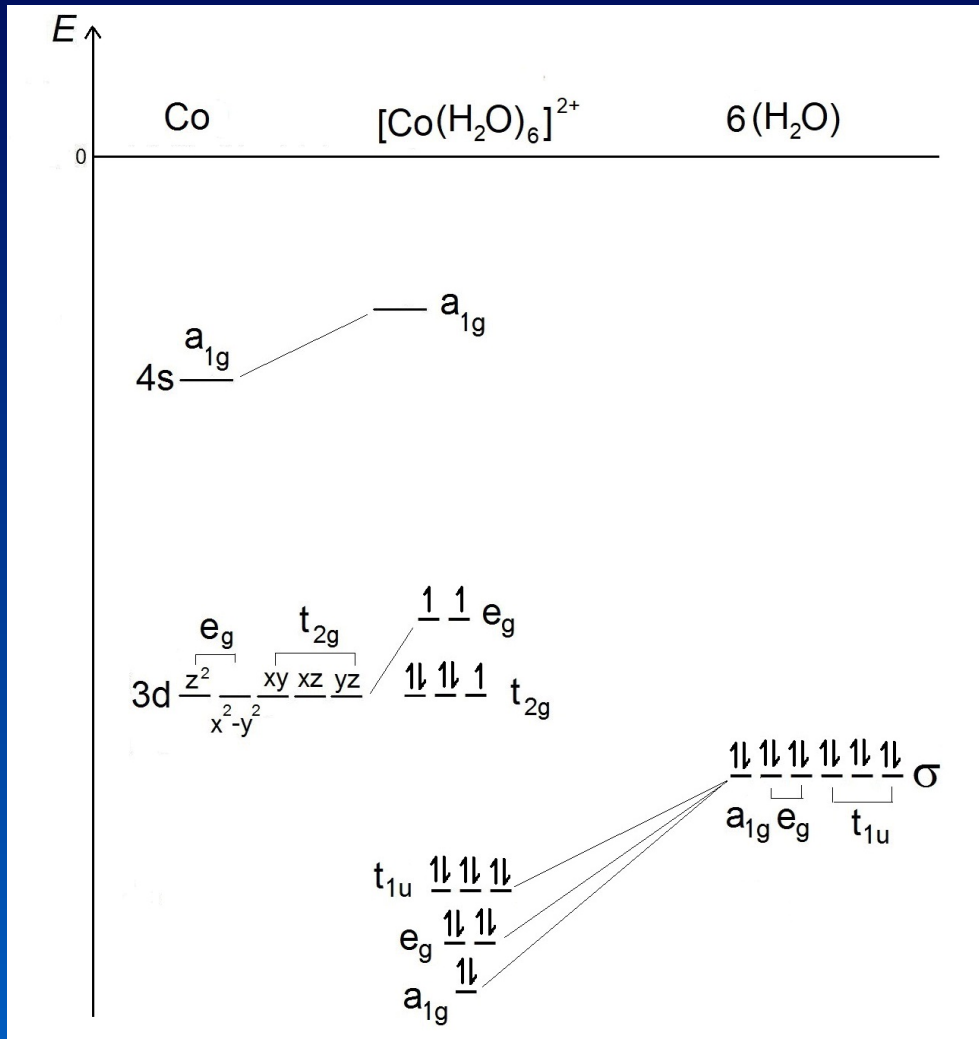


Com ligantes  $\pi$   
de campo forte



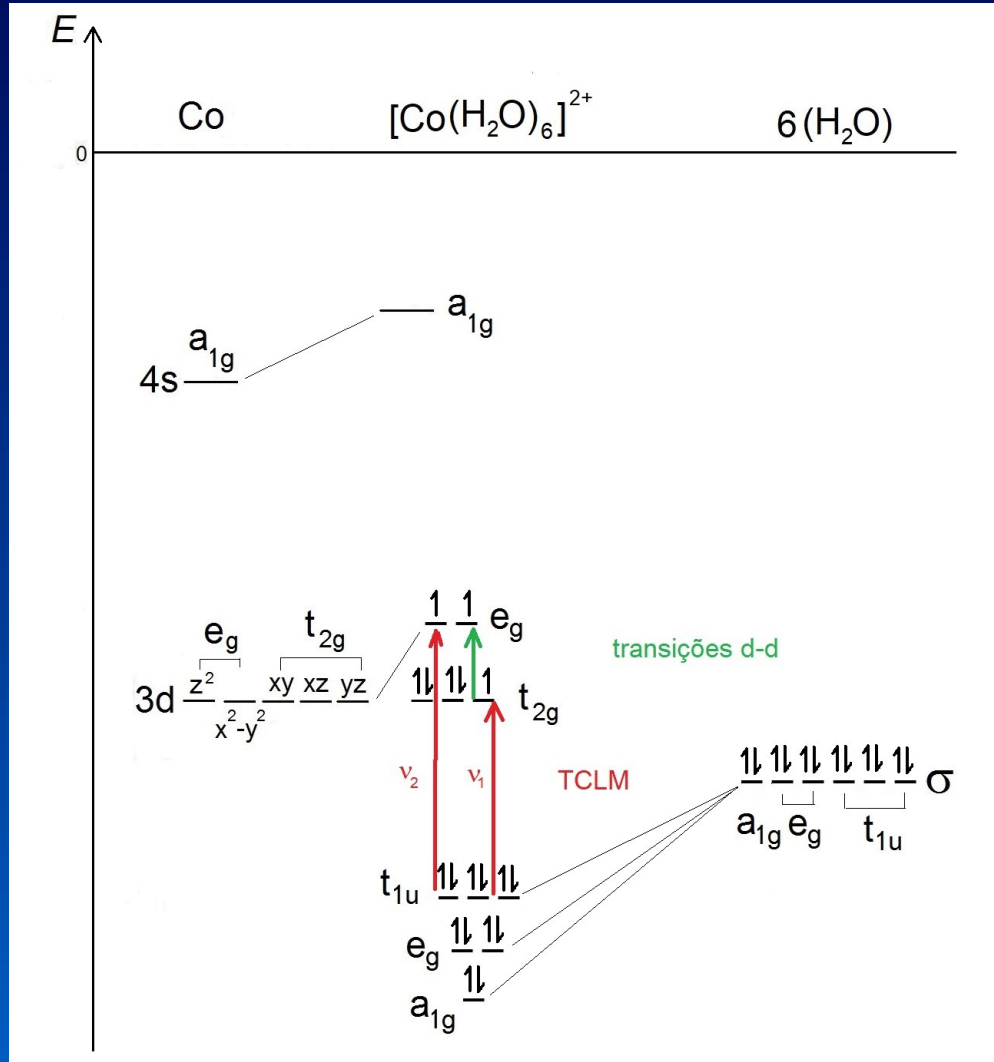
Sem ligantes  $\pi$

# Transições eletrônicas e espectroscopia



Sem ligantes  $\pi$

# Transições eletrônicas e espectroscopia



Sem ligantes  $\pi$



FIM